Synthesis, surface modifications, and biomedical applications of carbon nanofibers: Electrospun vs Vapor-grown carbon nanofibers

Samaneh Keshavarz¹, Oseweuba Valentine Okoro², Masoud Hamidi^{1,2}, Hossein Derakhshankhah³, Mehdi Azizi⁴, Seyed Mohammad Nabavi⁵, Shayan Gholizadeh⁵, Seyed Mohammad Amini⁷, Amin Shavandi², Rafael Luque⁸*and Hadi Samadian⁴*

¹ Department of Medical Biotechnology, Faculty of Paramedicine, Guilan University of Medical Sciences, Rasht, Iran, <u>sama_6929@yahoo.com</u>

² Université libre de Bruxelles (ULB), École polytechnique de Bruxelles-BioMatter unit, Avenue F.D. Roosevelt,
 50 - CP 165/61, 1050 Brussels, Belgium, <u>oseweuba.okoro@ulb.be</u>; <u>masoud.hamidi@ulb.be</u>;
 <u>Amin.Shavandi@ulb.be</u>

³ Pharmaceutical Sciences Research Center, Health Institute, Kermanshah University of Medical Sciences, Kermanshah, Iran. <u>derakhshankhah.hossein@kums.ac.ir</u>

⁴ Dental Implants Research Center, Hamadan University of Medical Sciences, Hamadan, Iran

⁵ School of Food and Biological Engineering Jiangsu University, Zhenjiang, China. <u>Nabavi208@gmail.com</u>

⁶ Department of Biomedical Engineering, Rochester Institute of Technology, Rochester, NY, USA, <u>sg4290@rit.edu</u>

⁷ Radiation Biology Research Center, Iran University of Medical Sciences, Tehran, Iran, <u>mohammadamini86@gmail.com</u>

⁸ Departamento de Quimica Organica, Campus de Rabanales, Edificio Marie Curie (C-3), Ctra Nnal IV-A, Km 396, Cordoba, Spain, <u>rafael.luque@uco.es</u>

* Corresponding authors: Hadi Samadian and Rafael Luque

Email:

h-samadiyan@alumnus.tums.ac.ir (Hadi Samadian)

rafael.luque@uco.es (Rafael Luque)



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Abstract

Engineered nanostructures are promising materials with promising properties, enabled by precise design and fabrication, as well as size-dependent effects. Biomedical applications of nanomaterials in disease-specific prevention, diagnosis, treatment, and recovery monitoring require precise, specific, and sophisticated approaches to yield effective and long-lasting outcomes for patients. Carbon nanofibers (CNFs) have interesting properties, such as good mechanical strength, high electrical conductivity, and desirable morphological features. Broadly speaking, CNFs can be categorized as vapor-grown carbon nanofibers (VGCNFs) and carbonized CNFs (e.g., electrospun CNFs), which have distinct microstructure, morphologies, and physicochemical properties. Besides their physicochemical properties, VGCNFs and electrospun CNFs have distinct performances in biomedicine and have their own pros and cons. Some review papers in the literature summarize and discuss different types of CNFs and their performance in the industrial, energy, and composites area, but there is room for a comprehensive review paper dialing with the CNFs from a biomedical point of view. Therefore, various types of CNFs, their fabrication and surface modification methods, and their applications in the different branches of biomedical engineering will be discussed in this review.

Keywords: Carbon Nanofibers; Electrospinning; Chemical Vapor Deposition; Biomedical Applications

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1. Introduction

Engineered nanostructures are fascinating materials with promising properties, enabled by precise design and fabrication, as well as size-dependent effects. Material properties at nanoscale can be modulated with precision to control the bulk properties effectively. Due to these characteristics, nanomaterials are widely used as alternatives to conventional materials to confer new properties and to improve or change specific properties [1, 2]. Biomedical applications of nanomaterials in disease-specific prevention, diagnosis, treatment, and recovery monitoring require precise, specific, and sophisticated approaches to yield effective and long-lasting outcomes for patients [3, 4]. Engineered nanomaterials can satisfy these demands due to their precise design and fabrication as well as their effective integration with biological systems [5, 6].

Various nanostructures and nanomaterials have been used in different branches of material science and biomedical engineering over the last decades. Based on the shape and dimensions, nanomaterials are divided into zero-dimensional (nanoparticles), one-dimensional (nanowires, nanofibers, and nanotubes), and two-dimensional (nanolayers, coating, and thin films) [7, 8]. Onedimensional nanomaterials have attracted significant attention recently due to their high aspect ratio, good cohesion with extracellular matrix (ECM), and resemblance to ECM [9]. Among onedimensional candidates for biomedical applications, nanofibers have shown promising performances because we can fabricate them from a wide variety of substances and apply accessible surface modification chemistries [9-11].

Carbon nanofibers (CNFs) have interesting properties, such as good mechanical strength and high electrical conductivity, along with desirable morphological features [12, 13]. Due to their high aspect ratio, it is possible to synthesize homogenous nanocomposites with a low amount (below 0.1 vol.%) of CNFs [12]. The large surface area can be applied to adsorb various sensing and

therapeutic agents in the diagnosis and therapy concepts, respectively. The electrical conductivity and mechanical properties of CNFs can accommodate the fabrication of sophisticated biosensors and electroconductive scaffolds with structural integrity for tissue engineering applications [14, 15]. Although CNFs have lower crystallinity, electrical conductivity, mechanical strength, and smaller surface areas than carbon nanotubes (CNTs), they have been shown to offer lower toxicity, fabrication costs, and simple dispersion.

Moreover, the defects in the CNFs structures can serve as active sites for reactions and bioconjugations [16, 17]. Depending on the fabrication method and the stacking manner of graphene layers, CNFs can have various forms, morphologies, and microstructures, as shown in **Fig. 1**. Broadly speaking, CNFs can be categorized as vapor-grown carbon nanofibers (VGCNFs) and electrospun CNFs, which have distinct microstructure, morphologies, and physicochemical properties [12]. VGCNFs are synthesized based on the catalysis methods and have more similarities to CNTs in terms of microstructure, crystallinity, mechanical and electrical properties. On the other hand, spun CNFs have an amorphous microstructure and lower mechanical properties, and electrical conductivity [18, 19].

Apart from their physicochemical properties, VGCNFs and spun CNFs have distinct performances in biomedicine and have their own pros and cons. Some review papers in the literature summarize and discuss different types of CNFs and their performance in the industrial, energy, and composites area [20-22], but there is room for a comprehensive review paper dialing with the CNFs from a biomedical point of view. Therefore, various types of CNFs, their fabrication and surface modification methods, and their applications in the different branches of biomedical engineering will be discussed in this review.



Fig. 1. Schematic structure of (a) graphene platelets, (b) graphene fishbones, (c) graphene ribbons and (d) cup-stacked CNFs and (e) amorphous CNF without graphene layers. Reproduced with permission from Ref. [23]

2. Carbon nanofibers

2.1. Synthesis

CNFs fabrication techniques can be classified into chemical vapor deposition (CVD)-based and carbonization of precursor nanofibers. The precursors nanofibers can be fabricated by template-based and spinning-based methods (force spinning, melt spinning, and electrospinning) following the carbonization process through the heat treatment. Moreover, naturally occurring fibers from cellulose, chitin, lignin, and chitosan can be converted to CNFs through the direct carbonization process [24] [25]. Fabrication of CNFs using CVD has a long story dating back to 1889 and the

synthesis of CNFs from hydrocarbon gas on metal crucibles. These nanomaterials are also called vapor-grown carbon fibers (VGCFs). Since then, various studies have been conducted to develop and optimize the synthesis method, and different hydrocarbons, such as carbon monoxide, benzene, acetylene, methane, and ethane, as the precursors and iron, cobalt, and nickel, as the catalysts have been evaluated [26]. In this method, the decomposition of hydrocarbon molecules on the metal catalyst takes place at a temperature range of 700–1200 °K and CNFs form through the diffusion of the carbon atoms (**Fig. 2**). It has been shown that the catalyst type determines the structure of the fabricated CNFs, for instance using Fe results in thick CNFs with a multi-walled CNT core, while Ni produces cup-stacked-type CNFs [23]. CNFs fabricated based on CVD methods possess excellent electrical conductivity due to the high graphitization temperature for various applications. Despite their fascinating properties, VGCFs require high production costs and hydrophobic surfaces, which has diminished their widespread applications [12].



Fig. 2. CNFs CVD instrumentation and growth mechanism.

The template-assisted fabrication technique uses porous membranes, hydrogel, or aerogel with well-defined porosity and channels as the template for the desired precursor materials. Using the porous membrane, the precursor materials pass through the channels via physical forces (e.g., piston pressure or ultrasonic waves) and fibers with diameters that match the pore sizes are obtained. The resulted fibers are converted to CNFs through the carbonization process using heat treatment. The template-based techniques are relatively simple and CNFs diameter depends on the membrane's pore size. For instance, Deeney et al. [27] used an Anodisc (anodized alumina) membrane with a 200 nm pore size as the template and beverage-related precursors to fabricate CNFs (**Fig. 3**). They reported that the obtained CNFs have diameter and length in accordance with membrane length and pore size. Despite its simplicity, it is not possible to fabricate fibers with long fiber lengths and micron-scale lengths are yielded.



Fig. 3. Schematic representation of template-assisted CNFs fabrication using membrane with defined pore size and length. Reproduced with permission from Ref [27].

Hydrogel/aerogels-based CNFs, also known as carbon gel, consist of three steps: cross-linking polymer chains, drying or solvent evaporation, and heat treatment or carbonization. The cross-linking can be conducted using physical, chemical, or enzymatic methods. The drying step is critical since it must preserve the porous architecture and pore size/morphology of the structure. Freeze drying and supercritical drying are commonly used drying technologies. Liang et al. [28] fabricated macroscopic monolithic carbon gel containing CNF based on ultrathin Te nanowires (TeNWs) dispersion in glucose solution. The resulting dispersion was converted to a robust monolithic gel-like Te/C structure using a heat treatment at 180 °C for 12–48 hours after the etching and washing to remove TeNWs (**Fig. 4a**). The resulted structure is carbon gen with uniform CNFs (**Fig. 4b**).



Fig. 4. Carbon gel fabrication overview. (a) schematic process representing the providing the sole solution and subsequent gel formation. (b) Macroscopic image of the resulting 12 L monolithic wet gel. (c) SEM micrograph with two different magnifications from the internal architecture of the carbon gel. Reproduced with permission from Ref [28]

The spinning methods are based on the extrusion of polymeric precursors from a spinneret under the influence of a force (Centrifugal force for the force spinning (**Fig 5a**), the electric field for electrospinning (**Fig 5b**), and hot air blowing for melt spinning (**Fig 5c**) [29]. The polymeric precursors can be dissolved in proper solvent/cosolvent or heated to melt in order to provide proper fluidity to pass through the spinneret. In the force spinning, the molten or solution of polymer pass through orifices under centrifugal force, and micro/nanofibers will form depending on the process parameters, such as rotational speed of the spinneret, orifices diameter and shape, collection system, and temperature (**Table 1**) [30]. In melt blowing spinning, the molten polymer extrudes from an orifice and is blown into continuous fibers by hot and high-velocity air and collected on a rotary drum. The morphology and diameter of the fibers depend on the airflow, air temperature, extruder morphology and temperature, and polymer feeding rate [31, 32].

Unlike mechanically-driven spinning, a high voltage electric field is the electrospinning method's main driving force of micro/nanofiber formation. The polymer solution or molten is charged using the applied electric field and is fed through the nozzle, pass through the spinning zone, and collected on the collector **Fig. 5b**). This technique is more sophisticated and effective in fabricating nanofibers, specially CNFs, with tuned diameter, morphology, and architecture. Moreover, it is possible to fabricate more complicated CNFs, such as hollow CNFs and aligned CNFs, using

electrospinning. The fabricated nanofibers' morphology, diameter, and architecture depend on various polymer, process, and environmental parameters.



Fig. 5. Various spinning-based CNFs fabricating methods. (a) Force spinning, (b) Electrospinning, and (c) Melt blown spinning.

In the carbonization methods, the type of the polymeric precursor has determinant effects, along with the process parameters. Various types of polymers can be utilized to fabricate CNFs (**Table 1**). Some of these polymers (known as host precursors) are used as the carbon source and converted to CNFs using the proper heat treatment, such as polyacrylonitrile (PAN) and polyimide (PI); on

the other hand, some other (known as guest precursors) are removed through the heat treatment(s) and leave special features in the finished CNFs, such as polystyrene (PS) and polymethyl methacrylate (PMMA). PAN has been extensively used due to its relatively high carbon yield (>50%), good processability, spinnability, and fabulous mechanical properties.

Table 1. Polymers used in electrospun carbon nanofibers fabrication as the host precursors or guest

 precursors

Precursor s	Chemical structure	Electrospinn ing condition	Functionalities	Pros	Cons
PAN	$\begin{array}{c c} H_2 & H_2 \\ & &$	4–10 wt.% in DMF, 10– 30 kV	Carbon source for CNFs and their hybrids	High carbon yield (>50%) and spinnability	Typically only DMF is used as solvent
Pitch	-	30–40 wt.% THF/DMF, 18– 25 kV	Carbon source for CNFs	High carbon yield (60% at 1000 °C)	Low spinnability
Lignin		20–35 wt.% in DMF/water, 6–26 kV	Carbon source for CNFs	Large surface area	Large diameter; low carbon yield of 20–40%

Polyimide resin	H H H H H H H H H H	10–20 wt.% in N-methyl, Pyrrolidone THF/metha ol/DMAc, 8–25 kV	Carbon source for CNFs, separator for LIB	High conductivity of asprepared CNFs	Complex fabrication process
PVDF	$ \begin{array}{c c} \mathbf{F} & \mathbf{H} \\ \hline \mathbf{C} & \mathbf{C} \\ \hline \mathbf{C} & \mathbf{C} \\ \hline \mathbf{F} & \mathbf{H} \\ \mathbf{n} \end{array} $	7–25 wt.% in DMAc/DMF / acetone, 15–20 kV	Separator and gel electrolyte for LIB	High mechanical properties	Require Dehydrofluorina ion for carbonization
PVP		4–10 wt.% in ethanol/DM F, 10–25 kV	Carbon source for metal oxide/CNF composites and template for neat metal oxide fibers	Soluble in various solvent	Low carbon yield of 15%
PVA		8-10 wt.% in water, 8–35 kV	Carbon source for metal oxide/CNF composites	Water soluble Good processabilit y	Low carbon yield of 3–10%
PMMA		8–10 wt.% in DMF, 10– 20 kV	Scaffold for neat metal oxide fibers and sacrificial phase for voids	Compatibl with other host polymer, such as PAN	Difficult to be fully removed, requiring 100 °C

PS	H H H H H H H H H H	5–35 wt.% in DMF/THF/ chloroform, 15–25 kV	Scaffold for neat metal oxide fibers an sacrificial phase for voids	Easy to be decomposed at -450 °C	Poor compatibility with host polymer, beads are easy to form
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Abbreviations: PAN: Polyacrylonitrile, DMF: Dimethylformamide, CNFs: Carbon nanofibers, THF: Tetrahydrofuran, LIBs: lithium-ion batteries, PVDF: Polyvinylidene difluoride, DMAc: Dimethylacetamide, PVP: Polyvinylpyrrolidone, PVA: Poly(vinyl alcohol), PMMA: Poly(methyl methacrylate), PS: Polystyrene

Fabricated electrospun PAN nanofibers are converted to CNFs using a two-step heat treatment, stabilization, and carbonization process. During the stabilization process, the cyclization and dehydrogenation process take place at 200-280 °C in air, which results in the formation of a ladder structure consisting of a series of connected C–N aromatic rings, as shown in Fig. 6. The stabilization is critical to preserve the nanofiber structure during the high-temperature carbonization [33]. The carbonization is usually carried out at a high temperature (more than 800 °C) in an inert atmosphere (nitrogen or argon), where the aromatic structure is grown and the heteroatoms like nitrogen are eliminated (**Fig. 6**) [34-36]. Other polymeric precursors can be converted to CNFs with a similar two-step heat treatment, although the exact processing conditions may differ.

Although the CVD-based synthesis methods provide CNFs with high graphitization and electrical conductivity, their production cost is remarkably higher than the spinning-based method. Moreover, using electrospinning, it is possible to synthesize CNFs with different morphology (solid, porous, and core/shell structured fibers) and architecture (random, aligned, and hybrid fibers) in a wide range of nanofibers diameter (10 nm to 10 μ m). On the other hand, the electrospinning technique provides a mat/sheet of CNFs and must be crashed/powdered to be

applied in nanocomposite applications and it is hard to obtain well dispersed, fine, and discrete CNF. The CVD-based methods have advantages in nanocomposite applications. For biomedical applications, the electrospun CNFs have priority over the CVD-based CNFs since they have lower toxicity issues. The possible toxicity of CVD-based CNFs can be attributed to the usage of metallic catalysts, which are not required in spinning-based methods.



Carbonaized strucutre

Fig. 6. Structural changes of PAN during the heat treatment process.

2.2. Surface modifications

Improving the surface properties of CNFs, such as colloidal stabilization, constitutes an important research interest due to the propensity of CNFs exist in aggregations, thus limiting their tissue engineering applications i.e., in polymer reinforcement [37]. Furthermore, the absence of sufficiently polar functional groups on the surface of CNFs may lead to weak interface bonding strength with polymers when employed in composite fabrication [38]. Klein et al. [39] also emphasized that improving the surface properties of CNF is crucial since these surface properties tend to dominate at the nanoscale because of the high surface-to-volume ratio of the CNF. Additionally, surface modification is necessary to enhance the interfacial adhesion and dispersion state of CNFs in polymer matrixes [40]. In composite fabrication, the surface modification of CNFs also can improve the mechanical properties of polymer-based composites. This is because the effective stress transfer from the polymer matrices to CNFs is optimized due to the presence of new covalent bonds between polymer matrices and the modified CNF surface [41]. CNF surface modification may also be employed to enhance its wettability [42]. In all cases of surface modification of CNFs, the objective is to modify surface chemistry as a pathway to enhance either the cell attachment sites, surface specificity, selectivity, or surface stability, depending on the CNF application interest (i.e., biological, composite applications, etc.) [39]. It must be emphasized that the modification of the CNF mainly serves to change the surface of the CNF without altering the fiber morphology [43]. Recognizing the importance of surface modification of CNFs for improving their properties, several approaches have been explored in the literature and are discussed in the subsequent sections (Fig. 7).



Fig. 7. Routine surface modification methods of carbon nanofibers

2.2.1 Thermal treatment

The CNF surface modification via thermal treatment for the addition of oxygenated functional groups may also be undertaken. For instance, in the study by Smolka et al. [44], the surface properties of CNFs were improved via oxidation in the air (i.e., dry oxidation) at the temperature of 800 °C while imposing a vacuum pressure. This study showed that the oxidation of the CNF surface led to improvements in CNF hydrophilicity and decreased the carbon content at the surface

and at near-surface sites for enhanced biocompatibility and improved electrical conduction. Similarly, surface activation of CNF via surface modification was also demonstrated by Liu et al. [45]. The study showed that oxidation in atmospheres containing 5 % v/v O₂ and Argon facilitated the introduction of C-O and O-H functional groups on the surface of the CNF. Such dry oxidation for the surface modification of CNF may also be achieved via ozonation [46]. This ozonation also facilitates the removal of residual amorphous carbons on the surface of the CNF while simultaneously leading to the generation and attachment of oxygenated functional groups, such as carboxylic, alcoholic, etc., via the 1,3 dipolar cycloaddition of ozone across double bonds [46]. Thermal treatments of CNFs may also be employed in introducing amine functionality to the surface of CNFs via a hydrothermal process[47].

For instance, Tshwenya et al. [47] showed that at a temperature of 180 °C for 6 h CNF could be modified by using 3 poly(propylene imine) as the amine group source. The now modified CNF, containing the amine functional groups was shown to demonstrate anionic diode behavior rather than the cationic diode behavior that characterizes pristine CNFs when immersed in an aqueous electrolyte solution [47]. This observation indicated that the charge properties of CNF may also be modified via surface modification techniques.

In some cases, surface modification via oxidation-based thermal treatments may be integrated with other surface modification strategies to introduce unique properties to the CNF surface, such as enhanced heavy metal ion adsorption from aqueous solutions [48]. For instance, in the study by Thamer et al. [48], the covalent bonding of melamine, and poly(m-phenylene diamine) with oxidized CNFs (O-CNFs) was employed to produce melamine-functionalized (m-O-CNF) and poly(m-phenylene diamine)-functionalized (p-O- CNF) CNFs, respectively. The study demonstrated that the treatment of the O-CNF with melamine and poly(m-phenylene diamine) led

to a ~ two-fold and ~4 fold increase in the adsorption capacity of the o-CNF. However, the functionalization of the CNFs surface via the addition of oxygen groups was also reported to impact their electronic conductivity, with higher levels of oxidation correlating with lower electronic conductivity [49]. These observations suggest that such oxidation-based surface modifications must be carefully modulated to avoid compromising CNF structural integrity and conductive properties.

2.2.2 Acid treatment

Surface modification of CNFs may also be achieved using oxidizing acids (i.e., wet oxidation) Klein, Melechko [39]. The surface modification using acids is also referred to as "Wet etch treatment" Klein, Melechko [39]. Such acid treatments typically utilize acids such as nitric acid and sulfuric acid to introduce oxygenated functional groups to the surface of CNFs. For instance, in the study by Din et al. [50], surface modification of CNFs was achieved via treatment using nitric acid solution [50]. The study showed that after the surface modification of CNFs via nitric acid treatment, C-OH and C=O functionalities were detected on the surface. Notably, oxidation approaches may facilitate the addition of several oxygenated functional groups to the CNF surface, as shown in **Fig. 7**.

Fig. 7 shows that oxidation may lead to the addition of C=O, OH, COOH, and COO⁻ functional groups to the surface of the CNF. Din et al. [50] also reported that the oxidation using nitric acid facilitated a ~36 % increase in the BET surface area of the CNFs while also facilitating the addition of functional groups to the CNF structure.

Another study also employed a similar strategy of utilizing nitric acid in the modification of the surface of the CNF as an approach to improve the resultant anti-wear properties when used in the

preparation of CNF/polytetrafluoroethylene (PTFE) composites. Nitric acid treatment was also shown to enhance the anti-wear property of the CNF (30 wt.%)/PTFE composite when subjected to a 200 N load [51], with Lim et al. [52] showing that such nitric acid treatments may also be used in the recovery of free edges that may have degraded when platelet-type, graphitized CNFs are subjected to the annealing process. Crucially, however, such acid treatments are capable of destroying the structural integrity of CNF [53]. This is because acid treatments may lead to excessive oxidations, leading to a severely oxidized surface laden with defects in the inner layers of CNFs, and destruction of its fibrous structure [49].

Notably, the acid treatment and thermal treatment approaches discussed above are based on the direct attachment of functional groups to the surface or the chemical modification of the surface using the oxygen-containing groups via sidewall defects [41]. Generally speaking, introducing oxygen-containing hydrophilic groups hydrophilic groups to CNF surfaces also facilitates their dispersion in polar solvents and enhances secondary reactions with other functional groups, thus leading to further alterations of CNF surface chemistry and properties [37, 54, 55]. Furthermore, it is important to acknowledge that although the well-known acid treatments can efficiently facilitate improved hydrophilicity of CNFs via the introduction of oxygen-containing groups, such as hydroxyl and carboxyl groups, on the CNF surface, it may not support dispersal in electrolyte or protein solutions. This is due to rapid sedimentation after a short period [56].

2.2.3. Plasma polymerization

Other CNF surface modification techniques such as plasma polymerization [57] may also be employed. This technique involves the formation of high-molecular-weight deposits (polymers) via activating monomers using an energetic plasma species [58]. The plasma source (i.e., UV) induces radical formation, which may recombine to form polymers [58]. The plasma polymerization technique was demonstrated in the study Pastine, Okawa [57], where surface modification via UV-triggered attachment of perfluoroarylazides was undertaken. It was shown that surface modification of CNF with hydrophilic, hydrophobic, or polymerizable small molecules led to surface functionalization such that superhydrophobic surfaces containing superhydrophilic regions were present. In another study by Guadalupe et al., [59], it was shown that CNF surface modification for enhanced hydrophobicity and dispersal properties was achievable via plasma polymerization using methyl methacrylate as the monomer. The effect of utilizing this modified CNF as a reinforcement for polymethyl methacrylate (PMMA) was also explored. This study showed that the adding up to 8 wt.% of the modified CNF facilitated an 8% and a 178% increase in the thermal stability and Young's modulus, respectively, compared to PMMA with no CNFs. A similar study showed that plasma surface modification of the CNF coupled with ultrasonication enhanced the dispersion property of CNFs in the CNF/polycarbonate composite [60]. In another study by Gao et al. [61], the surface of CNFs was modified via plasma polymerization for improved dispersion properties, as well as improved tensile strength and Young's modulus of composites of the polycarbonate matrix, compared to when pure CNFs, were dispersed in a polycarbonate matrix.

2.2.4. Surface modification via silane

In a study, surface modification of CNF was achieved via functionalization with 3glycidoxypropyltrimethoxysilane prior to its dispersion into epoxy resin. The study showed that the surface modification improved dispersion and interfacial interaction properties [37]. Surface modification may also be achieved via surface graft polymerization with acrylic acid using a KMnO₄/H₂SO₄ redox-induced system [62]. This approach facilitates a ~ two-fold increase in surface carboxylic acid groups for enhanced adhesion via stronger interfacial bonds when mixed with an epoxy matrix [62].

2.2.4. Chemical vapor deposition of thin film coatings

CVD is a technique that facilitates the deposition of solid material from a vapor via chemical reactions occurring on or in the vicinity of a normally heated surface and is generally classified based on the operating conditions and the physical characteristics of the vapor [63]. The CVD surface modification approach synthesizes thin films on the CNF surface by polymerizing vapor phase monomeric reactants and their subsequent deposition [64]. The properties of the films are typically tunable, thus permitting the development of CNF with unique characteristics [64]. According to Klein et al. study [39], CVD may be employed for film deposition on the surface of CNFs to impact different functionalities. The deposited films may be either dielectric, metals, metallic oxides, or polymers. For instance, in the study by Shi et al. [65], pyrrole was deposited as thin films on CNTs for surface modification. In another study [66], a silicon oxide film was deposited via CVD for surface modification by a novel self-aligned thin-film deposition technique such that the surface coating composition was stoichiometric SiO₂.

2.2.5. Non-covalent functionalization for CNF surface modification

The non-covalent functionalization approach involves CNFs surface modification using supramolecular complexations and is conducted through the Van der Waals forces, hydrogen bonds, electrostatic forces, and π -stacking interactions [67]. These non-covalent functionalizations may be referred to as 'non-invasive' as it does not destroy the structure of the CNFs and involve 'wrapping' CNF with another material [67]. Polymer wrapping is an example of such non-covalent functionalization (**Fig. 7**) [46]. The polymer wrapping facilitates the maintenance of the Sp² conjugated structure [46], with the dispersion of the CNFs in a given polymer now dependent on

the miscibility of the polymer chain employed in wrapping the CNFs and the polymer matrix [46]. Generally speaking, the effectiveness of the non-covalent functionalization surface modification is dependent on the strength of the bonds between the CNFs and the polymer wrapping that arises due to van der Waals forces and π - π stacking between CNFs and polymer chains containing aromatic rings, respectively [46]. For instance, non-covalent functionalization which involves supramolecular complexation and employs Van der Waals forces, hydrogen bonds, electrostatic force, and π -stacking interactions.

2.2.6 Electrochemical modification of CNF surface

The electrochemical surface modification approach involves the deposition of materials to decorate the surface of the CNFs, as shown in **Fig. 8a** [68]. In the study by Meng et al. [68], electrochemical modification of the CNFs surface was achieved electrochemically via a dynamic liquid membrane. This study demonstrated the possibility of introducing the functional groups of –OH and –COOH to the surface of the CNFs, thus effectively enhancing the CNFs' electrical conductivity and hydrophilicity. In another study, similar electrochemical treatments were employed to facilitate the alteration of the surface oxidation state via electron transfer. Such electron transfers enabled the functionalization of the surface for enhanced roughness, although unfavorable effects on the fiber strength were also reported [69].

Surface modification by introducing oxygenated functional groups (i.e. NiO) may also be possible via nickel surface coating, which increases surface polarity and surface free energy for improved fracture toughness and interfacial bonding when employed in composite systems based on the phenolic matrix [70].



Fig. 8. (a) Fundamentals of surface functionalization of CNFs via electrochemical approach (b) Illustration of mass transport during electrochemical surface modification of CNF electrode. (c) Highlighting the interactions at surface of the modified CNF. ([68] Copyright © (2020) American Chemical Society).

This section of this study has explored some of the major functionalization strategies for the surface modification of CNF and the associated improvement in their usefulness in tissue engineering applications. Several surface modification approaches, such as ozone, and plasma treatment, were discussed in this section. This section showed the viability of employing these

CNF functionalization techniques for the introduction of different functional groups such as hydroxyl, carboxylic, etc. on the CNF surface could alter essential properties (i.e., he BET area, elasticity, abrasion properties, etc.) of the materials when the modified CNF was introduced.

3. Biomedical applications

In recent years, electrospun nanofibers have emerged as attractive materials for numerous biomedical areas, including controlled drug delivery systems, biosensing, wound healing (as dressing), and tissue engineering (TE) [71, 72]. Fabricating the nanofibers via the electrospinning technique provides considerable advantages compared to other techniques (e.g., drawing, freezedrying, phase separation, template synthesis, and self-assembly) [73, 74]. High surface-to-volume ratio, narrower diameter, high porosity, enhanced mechanical strength, tuneability, simplicity, and economical properties highlight using the electrospinning method [75-78]. Furthermore, this method has the ability to control and optimize parameters such as tip collector distance, fiber direction, and voltage [75, 76]. In TE, the electrospun nanofibers may be used as scaffolds for mimicking the native nanoscale entities of the three-dimensional (3D) tissue extracellular matrix (ECM) to give rise to adhesion, proliferation, and differentiation of the cells and eventually regeneration of the damaged tissue [79]. Various natural (i.e., derived from natural sources) or synthetic scaffolds (i.e., synthesized from synthetic materials) have been electrospun by modulating the parameters in the electrospinning process [80]. Carbon-based nanomaterials (CNMs), including CNFs, CNTs, and graphene-based nanocarbons (the latter two are allotropes of carbon) [81, 82], have attracted much interest for biomedical applications, especially in TE, due to their unique electrical conductivity, chemical, thermal, mechanical, and structural properties [83, 84]. The electrospinning technique has also contributed to fabricating CNMs, including CNF,

which can be referred to as electrospun CNF (ECNF) [74]. In this regard, the current review gives an insight into the biomedical applications of ECNFs in biosensors, and TE for skin, bone, neural, and musculoskeletal tissues.

3.1. Biosensors

Biosensors are analytical tools with applications in several fields, such as biomedical, to detect the presence and concentration of an analyte (e.g., glucose, cancer biomarkers, drugs, metabolites, toxins, etc.) [85, 86]. These devices convert physiological or biological signals to electrical signals that represent the concentration of a given biochemical agent [87, 88]. The construction of biosensors is commonly composed of a biorecognition element (BRE) as a bioreceptor, a transducer (electrode), and a detector for processing to get a signal output [86, 89, 90]. Bioreceptors can be enzymes, antibodies, nucleic acids, aptamers, microorganisms, cells, tissues, or even synthetic molecules (e.g., molecularly imprinted polymers (MIPs) which are immobilized on transducer as depicted in **Fig. 9** [91-93]. Different transducer types include electrochemical, optical, piezoelectric, mechanical, acoustic, calorimetric, or magnetic. The most frequently used type is the electrochemical biosensors (comprising amperometry/voltammetry, potentiometry, and impedometry techniques) owing to cost-effectiveness, simplicity, and sensitivity [86, 91]. Cyclic voltammetry (CV), differential pulse voltammetry (DPV), linear sweep voltammetry (LSV), and square wave voltammetry (SWV) are among the electrochemical amperometric biosensors [94].



Fig. 9. Schematic illustration of bioreceptors and transducer structures in a biosensor.

The sensitivity and detection limit of the biosensor are essentially determined by bioreceptors and transducers [95]. Different samples from humans (e.g., blood, urine, saliva, sweat, etc.), food (e.g., milk, juice, etc.), and environment (e.g., air, water, soil, etc.) can be used in biosensors [95, 96]. Biosensors can provide simple, specific, reliable, accurate, cost-effective, and real-time results [96]. CNMs such as ECNFs could be applied in fabricating the biosensors to improve their performance (e.g., sensitivity) [97-99] due to being functionalizable, biocompatible, sensitive, and

cost-effective, and possessing electrocatalytic activities, mechanical strength, and high surface area to the volume [85, 100-102]. Adabi et al. [103] demonstrated that the diameter of ECNFs ranging from 75 to 80 nm has desired performance and conductivity and can serve as nanoelectrodes.

ECNF-based biosensors have been applied in the electrochemical detection of various biomedical agents such as H₂O₂, glucose, dopamine (DA), drugs, etc. An H₂O₂ biosensor was created from the carboxylic acid group-functionalized ECNF (FCNF) combined with hydroxyapatite nanostructures coated on a polished Au electrode and subsequent immobilization of cytochrome c on its surface (Cyt c/FCNFs-HA/Au electrode). This composite has exhibited a good electrocatalytic activity (at an applied potential of -0.05 V), a fast response time (5 s), and a limit of detection (LOD) of 0.3 µM [100]. A wide linear range has been achieved by diverse H₂O₂ biosensors such as electrospun nanoporous CNF-platinum nanoparticle/glassy carbon electrode (GCE) (with 2 linear regions: 10 µM-9.38 mM and 9.38-74.38 mM) [101], electrospun cobalt nanoparticles/CNFGCE (up to 5 mM) [104], and electrospun nitrogen-doped CNFs (0.01–0.71 mM) [105]. A glucose biosensor as electrospun nickel-cobalt(II) oxide (Ni-CoO) loaded CNF indicated a wide linear range (0.25-600 μ M), a low LOD (0.03 μ M), and good stability [106]. In a study by Wang et al. [107], an enzyme-based biosensor utilizing FCNF (CNFs were prepared by electrospinning of PAN) was developed for detecting glucose. In their study, glucose oxidase (GOD) was immobilized on a Prussian blue-FCNF modified GCE (PB-FCNF/GCE) and represented a rapid response time (5 s), a low LOD (0.5 µM), a wide linear range (0.02–12 mM), high sensitivity (35.94 μ A cm⁻² mM⁻¹), and good selectivity towards glucose In a related study by Li et al. [98], an enzyme-based biosensor has been designed by employing ECNFs (laccase-Nafion-ECNFs on the GCE) for the determination of catechol. This composition displayed good

repeatability, reproducibility, stability, and selectivity. Another enzyme-based biosensor (electrospun Cu/CNFs/laccase(Lac)/Nafion/GCE) has been developed by Fu et al. [108] for catechol biosensing. Their investigation also exhibited good repeatability, reproducibility, stability, and selectivity with a rapid response (within 5 s), a wide linear range (9.95 $\times 10^{-6}$ -9.76 $\times 10^{-3}$ M), and a low LOD (1.18 μ M).

An electrospun Pd nanoparticle-loaded CNFs (Pd/CNFs) nanocomposite for detecting DA, uric acid (UA), and ascorbic acid (AA) demonstrated good sensitivity, selectivity, stability, and reproducibility [109]. A fast response, high sensitivity, wide linear range, low LOD, good stability and selectivity were obtained by silver-platinum/electrospun nanoporous CNFs-modified GCE (Ag-Pt/pCNF/GCE), another biosensor for determining DA [102]. It is demonstrated that an ECNF-modified carbon paste electrode (CPE) has led to a low LOD of 0.02-11.47 µM, a wide linear range of 20 nM, as well as a high sensitivity of 10 nA µM⁻¹ towards dihydronicotinamide adenine dinucleotide (NADH) [110]. Electrospun CPE/Pd/CNF/methyl (trioctyl)ammonium bis(trifluoromethylsulfonyl)imide cobalt/Nafion [111] and ECNFs/screen printed electrode (SPE) [112] are ECNF-based biosensors established for the determination of the pemetrexed (PTX), an anti-cancer drug and tramadol, an opioid pain medication, respectively. These biosensors generated good stability with high sensitivity and selectivity. For detecting xanthine (Xa), a purine base, an ECNF-modified CPE was fabricated by Tang et al. [113]. This biosensor's linear range and LOD were 0.03–21.19 µM and 20nM, respectively. In addition, ECNF-CPE presented good stability, selectivity, and sensitivity. Cai et al. [114] developed a pressure sensor (FeOCN) based on the ECNF consisting of ferrosoferric oxide (Fe3O4). A wide working range (0-4.9 kPa), a rapid response time (0.43 s), an ultralow LOD (6 Pa.), a high sensitivity (0.545 kPa⁻¹), and good stability were acquired in this work.

Table 2 summarizes different characteristics related to various carbon-based biosensors such as different electrochemical detection methods, selectivity, linear range (LR), response time, the limit of detection (LOD), and sensitivity for the detection of biomarker/biomedical agents. An ideal biosensor represents the rapid response, low detection limit, and high sensitivity and selectivity for the intended target [86].

Biomarker/	Biosensor composition	Method(s)	Selectivit	LR	Respons	LOD	Sensitivity	Real	Ke	y findings	Ref.
biomedical agents			У		e Time (s)			sample			
Ascorbic acid (AA)	Electrospun Pd/CNFs	CV, DPV	DA, UA, AA	0.05–4 mM	-	15 μΜ	-	Urine	Ex Hi Go Ac rep	ccellent selectivity, igh sensitivity, ood stability, cceptable producibility	[109]
Carcinoembryoni c antigen (CEA)	CNTs/PDDA/HRP/ConA/HRP- Ab2	CV, DPV, EIS	CEA	0.05–5 to 5–200 ng mL ⁻¹	-	0.018 ng mL ⁻¹	-	Serum	 Hi Lo Lo ma Co 	igh sensitivity, ow detection limit, ong-term bioactivity aintenance, ost-effectiveness	[115]
Catechol	Laccase–Nafion–ECNFs on GCE	CV, chronoamperometr y	Catechol	1–1310 μM	2	0.63 μΜ	41 μA mM ⁻¹	Water	 Go rep sta Ex 	bod repeatability, producibility, and ability, acellent selectivity	[98]
Catechol	Electrospun Cu/CNFs/laccase(Lac)/Nafion/GC E	CV	Catechol	9.95 ×10 ⁻⁶ - 9.76 ×10 ⁻³ M	Within 5	1.18 μM	33.1 μA mM ⁻¹	-	 Go Ra W Lo Go rej sta se 	bod biocompatibility, apid response, 'ide linear range, bw detection limit, bod repeatability, producibility, ability, and lectivity	[108]
Dopamine (DA)	Electrospun Pd/CNFs	CV, DPV	DA, UA, AA	0.5–160 μM	-	0.2 μΜ	-	Urine	Ex Hi Go Ac rep	ccellent selectivity, igh sensitivity, ood stability, cceptable producibility	[109]
Dopamine	ZnO NF/CF	CV, DPV	DA	6–20 μM	-	0.402 μM	-	-	• Hi sel • W	igh sensitivity, and lectivity, 'ide linear range	[116]
Dopamine	Electrospun Ag-Pt/pCNF/GCE	CV, DPV	DA	10-500 μΜ	-	0.11 μΜ	-	-	 Fa Hi W Lo Go Ex 	ast response, igh sensitivity, i'de linear range, ow detection limit, ood stability, scellent selectivity	[102]
Glucose	NiMoO ₄ /CNF membrane	CV, EIS	Glucose	0.0003– 4.5 mM	5	50 nM	301.77 μA cm ⁻² mM ⁻¹	Human blood serum	 Exant Lo W 	ccellent sensitivity, d selectivity, ow detection limit, fide linear range,	[117]

 Table 2. Summary of different characteristics
 related to various carbon-based biosensors for the detection of biomarker/biomedical agents

									•	Reliable, repeatable, and stable, High specificity	
Glucose	Electrospun GOD/PB-FCNF/GCE	CV, chronoamperometr y	Glucose	0.02–12 mM	5	0.5 μΜ	35.94 μA cm ⁻² mM ⁻¹	-	•	Rapid response, Low detection limit, Wide linear range, High sensitivity, Good stability, repeatability, and selectivity	[107]
Glucose	PdNPs-ERGO-GOD/GCE	CV, chronoamperometr y	Glucose	25 μM- 4.9 mM	Within 5	0.56 μM	-	Human blood serum	•	Wide linear range, Low detection limit, Good reproducibility, Long-term stability, High selectivity	[118]
Glucose	Cu ₂ O NCs/r-GO/GCE	LSV	Glucose	5.0- 9595 μ Μ	Less than 3	1.0 μM	23.058 µA mM [−] 1	Human blood	•	Wide linear range, Low detection limit, High sensitivity, and selectivity	[119]
Glucose	Electrospun Ni-CoO/CNF	CV, EIS	-	0.25- 600 μM	-	0.03 μM	-	Serum	•	Wide linear range, Low detection limit, Good stability	[106]
Glucose	NiCo/TiO2/CNFAs	CV, EIS	Glucose	1- 7658 μ Μ	Within 5	0.6 µM	975.3 μA cm ⁻² mM ⁻¹	Human blood	•	Wide linear range, Low detection limitation, Excellent stability, high sensitivity, and selectivity	[120]
Glucose	Ni-NaA/CPE	CV, DPV, chronoamperometr y	Glucose	0.04– 6.54 mM	Less than 60	1.98×10 -3 mM	225.8 μA cm ⁻² mM ⁻¹	Blood serum	•	High selectivity and sensitivity, Good reproducibility	[121]
H_2O_2	Electrospun Cyt c/FCNFs-HA/Au	CV, chronoamperometr y	H_2O_2	2.0 μM- 8.7 mM	5	0.3 μΜ	-	Blood	•	Good electrocatalytic activity, Fast response,	[100]
H ₂ O ₂	Electrospun nanoporous CNF- PtNP/GCE	CV, chronoamperometr y	H ₂ O ₂	2 linear regions: 10 μM- 9.38 mM and 9.38- 74.38 mM	-	1.9 μΜ	-	-	•	Wide linear range, Low detection limitation, High selectivity, and sensitivity	[101]

H_2O_2	Electrospun Co-NC/CNFGCE	CV, chronoamperometr y	H ₂ O ₂	Up to 5 mM	-	10 μM	300 μA cm ⁻² mM ⁻¹	Juice, milk	•	Wide linear range, High selectivity	[104]
H_2O_2	Electrospun N-doped CNFs	ĊV	-	0.01– 0.71 mM	Within 30	0.62 μM	357 μA cm ⁻² mM ⁻¹	Milk	•	High sensitivity, Low detection limit, Wide linear range	[105]
NADH	Electrospun CNF-CPE	CV, DPV, EIS	NADH, AA	0.02- 11.47 μM	5	20nM	10 nA μM ⁻¹	-	•	Low detection limit, Wide linear range, Good selectivity	[110]
Pemetrexed (PTX) *	Electrospun CPE/Pd/ CNF/[M3OA]+[NTF2]–/Nafion	SWV, CV	PTX	1.00– 35.0 nM	-	0.33 nM	-	Cancero us human plasma, pharma ceutical, health urine samples	•	High sensitivity, selectivity, rapidity, and simplicity Good stability	[111]
Pressure	Electrospun FeOCN	-	-	0–4.9 kPa	0.43	6 Pa.	0.545 kPa ⁻¹	Wrist pulse, phonati on, breathin g, finger bending	•	Wide working range, High sensitivity, Ultralow detection limit Rapid response time, Good stability, High hydrophobicity, Excellent flexibility	[114]
Tramadol	Electrospun CNFs/SPE	CV, SWV	Tramadol	0.05– 100 nM	-	0.016 nM	-	Urine	•	Excellent selectivity, High sensitivity, Good linearity, repeatability, reproducibility, and stability, Rapid	[112]
Uric acid (UA)	Electrospun Pd/CNFs	CV, DPV	DA, UA, AA	2–200 μM	-	0.7 μΜ	-	Urine	• • •	Excellent selectivity, High sensitivity, Good stability, Acceptable reproducibility	[109]
Xanthine (Xa)	Electrospun CNF-CPE	CV, DPV, EIS	Xa	0.03– 21.19 μΜ	-	20nM	-	Human urine, fish	• • •	Rapid response, Low detection limit, Wide linear range, Good stability, selectivity, and sensitivity	[113]

Abbreviations: CV: cyclic voltammetry; DPV: differential pulse voltammetry; EIS: electrochemical impedance spectroscopy; SWV: square wave voltammetry; LSV: linear sweep voltammetry; LR: Linear range; LOD: Limit of detection; Pd/CNF: Palladium/Carbon nanofiber; CNTs/PDDA/HRP/ConA/HRP-Ab2: Carbon nanotubes/poly(diallyldimethylammonium chloride)/horseradish peroxidaselabeled/concanavalin/horseradish peroxidase-labeled carcinoembryonic secondary antibodies; ECNF: Electrospun CNF; GCE: Glassy carbon electrode; ZnO NF/CF: Zinc oxide nanofiber/carbon fiber; Ag-Pt/pCNFs: Silver-Platinum/Electrospun nanoporous CNFs; NiMoO4: Nickel molybdate; GOD/PB–FCNF: Glucose oxidase/Prussian Blue-Functionalized CNF; PdNPs-ERGO: Pd Nanoparticles-Electrochemically Reduced Glucose oxidase; Cu2O NCs/r-GO: Copper(I) oxide Nanoclusters/reduced-Graphene oxide; Ni-CoO: Nickel-Cobalt(II) oxide; NiCo/TiO2/C

3.2. Tissue engineering

TE aims to regenerate diseased or injured tissues and organs via three main factors known as "the TE triad" including cells (e.g., stem cells or differentiated cells), inductive agents (e.g., growth factors or small molecules), and scaffolds (natural, or synthetic) [122-124]. It is essential to fabricate scaffolds as ECM without side effects, cytotoxicity, or immunological rejection [125-127]. The major roles of ECM components are to provide the growth factors, induce signaling pathways, and direct cell-ECM communication [72]. A wide range of natural and synthetic nanofiber materials have been investigated in the TE field for different tissue types such as skin, bone, musculoskeletal, neural, and so forth [128-130]. Naskar et al. [131] fabricated VGCNFs reinforced nonmulberry silk protein fibroin films via solvent evaporation technique as an option for tissue regenerations (e.g., muscle, bone, and nerve tissue). They demonstrated that VGCNFs endow the composite films with outstanding electrical conductivity, tensile modulus, hydrophilicity, and thermal stability, providing better fibroblast cell growth and proliferation. Additionally, many studies have been conducted to investigate the potential application of electrospun scaffolds in this field [76]. For instance, as mentioned, ECNFs owing to significant mechanical properties, are promising tools for bone TE [132, 133].

3.2.1. Skin tissue engineering/wound healing

The skin is the body's largest organ, and it protects against foreign substances and organisms [134]. Serious wounds for any reason (e.g., burns, diabetes, infections, etc.) could result in loss of the reparative and regenerative capabilities of the skin cells [135]. Skin TE provides replacement therapy and rapid healing via ideal wound dressing, limiting the risk of infections that could negatively impact skin tissue regeneration [136]. Several natural (e.g. collagen, elastin, fibronectin,

chitosan, etc.) and synthetic ((e.g. poly(ethylene glycol) (PEG), poly(lactic acid) (PLA), polyglycolide (PGA), poly(lactic-co-glycolic acid) (PLGA), poly-ɛ-caprolactone (PCL), etc.) polymers have been utilized for the skin TE [134, 137]. Salesa et al. [138] developed calcium alginate/VGCNFs composite with antibacterial activity for the first time to address the antibiotic resistance problems. This composite is not only cost-effective but also shows no cytotoxicity on human keratinocytes (HaCaT cells). The same group incorporated 1D hydrophobic VGCNFs and 2D hydrophilic GO nanosheets into alginate films to assess their effects on HaCaT cell adhesion. They demonstrated that this composite showed no cytotoxicity but did not improve cell adhesion [139]. In another study, copper and zinc distributed in VGCNFs and grown on an activated carbon fiber substrate (Cu-Zn-ACF/CNFs) inhibited the growth of the Gram-negative bacteria (Escherichia coli, Gram-positive Staphylococcus aureus, and Methicillin Resistance Staphylococcus aureus), with high efficiency. In this study, this composite was suggested as an antibiotic wound dressing [140]. Li et al. [141] also designed a potential hemostatic patch (superhydrophobic VGCNFs coating). This VGCNFs gauze promoted rapid fibrin growth and clotting, and reduced bacteria attachment in an *in vitro* study.

3.2.2. Bone tissue engineering

Bone TE (BTE) is an emerging therapeutic approach to regenerate and repair bone defects [142]. Principally, in the case of critical-sized bone defects in which the bone cannot heal spontaneously, TE might offer an effective treatment strategy [143, 144]. Introducing CNMs into biomedical applications has found prominent opportunities for BTE [145]. CNTs/CNFs are considered reinforcing agents for polymers such as PCL, PLA, and so on to enhance or improve their mechanical, biocompatibility, and cellular responses and resemble the nanoscale architecture of natural ECM as well [146].

CNFs have been shown to enhance the adhesion, and proliferation of osteoblasts, induce alkaline phosphatase (ALP), and Ca^{2+} secretions (osteoconductivity effects), and possess excellent mechanical strength three times more than that of native bone tissue with Young's modulus of 2 TPa and the tensile strength similar to the bone [147-150]. Besides, these materials have unique chemical, electrical, morphological, and biological characteristics [143]. Zhou et al. [151] developed ECNFs through the carbonization of aligned electrospun PAN nanofibers bundles. They showed that with increasing the final carbonization temperature, the CNFs have been found more graphitic and structurally ordered, and anisotropic electrical conductivities. Yang et al. [152] fabricated electrospun PAN-based ECNFs containing bioactive glass nanoparticles (ECNFs/BG) by combining electrospinning and sol-gel techniques to evaluate their osteocompatibility. In this study, a sol-gel solution was produced by triethyl phosphate (TEP), calcium nitrate tetrahydrate $(Ca(NO_3)_2)$, and tetraethoxysilane (TEOS). Then, the resulting sol-gel solution with different compositions (68S, 77S, and 86S) was added to a PAN solution in N,N-dimethylformamide (DMF) to obtain BG solutions. These composites indicated an improved formation of apatite (in vitro biomineralization) and MC3T3-E1 osteoblasts proliferation. However, ECNF/BG(68s) was the best substrate for the adhesion and proliferation of MC3T3-E1 cells by having the highest silica solubility, and the ability to form apatite (Fig. 10). Remarkably, reinforced compressive strength and elastic modulus of 45S5 Bioglass scaffolds (ranging from 2.08 to 4.56 MPa and 111.50 to 266.59 MPa, respectively) synthesized by the sol-gel technique have been achieved by adding 0.25 wt.% MWCNTs [153].



Fig. 12. Well-attached on all the ECNF membranes (pure ECNF and different ECNF/BG composites) and spindle-shaped MC3T3-E1 cells along the fiber direction were observed by scanning electron microscope (SEM) on different days. Although the cell number increased with the culture time, ECNF/BG(68s) had the highest cell density and the largest spreading area. Reproduced with permission from Ref. [152]

Another study by Samadian et al. [154] exhibited that tailoring the mineralized electrospun PANbased ECNFs (M-ECNFs) by three artificial neural networks (ANN) models, each with different training algorithms for data analysis to fabricate electro-conductive scaffolds, possesses the potential application in BTE. The mentioned features are essential criteria for achieving an ideal bone regeneration scaffold. Notably, in contrast to CNTs, ECNFs display non-cytotoxic characteristics and could be considered safe and promising materials for bone, orthopedic, and dental TE and prosthetic, or implant applications [128, 147]. Wu et al. [155] synthesized 3D electrospun PAN-based ECNFs/hydroxyapatite (HAp) composites (**Fig. 13**) with strong interfacial bonding sites (strong interaction of HAp with the ECNFs through coordination bonds) and high mechanical strength that have potential in BTE. Furthermore, the carboxylic groups-activated ECNF surfaces made them hydrophilic.



Fig. 13. Synthesis of ECNFs/HAp composites scaffold for bone tissue engineering. (a) Schematic representation of the experiment, (b) SEM micrograph of pristine ECNFs after mineralization, (c) SEM micrograph of NaOH-treated ECNFs after mineralization, and (d) TEM micrograph of mineralized ECNFs. Reproduced with permission from Ref. [155]

Another 3-D CNF/HAp scaffold for bone tissue regeneration was produced via carbonization of bacterial cellulose (BC) nanofibers by Wan et al. [156]. In their work, nitric acid (HNO₃)-surface treated CNFs encouraged the mineralization (HAp nucleation and growth) and changed the morphology of HAp crystals developed on CNFs (rod-like vs. needle-like) (Fig. 14). Electroconductive ECNFs can also provide substrates for bone regeneration objectives. It is believed that the existence of external electrical stimulation (e.g., Direct current field (DCF), electromagnetic field (EMF), inductive coupling (IC), and capacitive coupling electrical field (CCEF)) for scaffolds is a potential osteogenic factor to regenerate bone. DCF is more popular because of its simplicity, and site-specificity. This factor can induce mineralization, proliferation, and differentiation of bone cells. In this regard, electro-conductive ECNFs-medicated DCF increased MG-63 cell growth and their osteogenic activity [147]. Since proliferation, differentiation, and mineralization of osteoblast cells could be encouraged by silica nanoparticles (SNPs), they are considered other osteoinductive agents [157]. ECNFs/SNPs have also been reported to improve attachment, viability, and proliferation of MG-63 cells in a related study [143]. Table 3 summarizes different carbon-based scaffolds applied for BTE.



Fig. 14. Transmission electron microscopy (TEM) images and selected area diffraction (SAED) patterns of various CNFs/HAp hybrids after 7 days of mineralization (a, b) as-prepared CNFs, (c, d) CNFs oxidized in 3 M HNO3, (inset of d) energy-dispersive X-ray spectroscopy (EDS) pattern of CNFs oxidized in 3 M HNO3, (e, f) CNFs oxidized in 10 M HNO3, and (inset of f) high magnification of f. Reproduced with permission from Ref. [156].

Scaffold composition	Cells (in vitro)	Key findings	Ref.
CNF compacts & PLGA casts of nanophase carbon fibers	Human osteoblasts	Selectively enhanced osteoblast adhesion	[158]
Electrospun ECNFs/HA	MG-63	 Enhanced mechanical strength by mineralization process (was not statistically significant (p < 0.1)) Biocompatible (24M-CNFs with negligible toxicity) 	[33]
Electro-conductive electrospun CNFs-medicated DCF	MG-63	 Increased cell growth (116.43 ± 4.76%, at 100 μA) Increased osteogenic activity 	[147]
ECNFs/SNPs	MG-63	Improved cell attachment, viability, and proliferation	[143]
Electro-conductive electrospun CNFs/Fe2O3	MG-63	 Cytocompatible Negligible toxicity (CNFs/Fe2O3 from PAN FeSO4·7H2O 15%) 	[133]
Electro-conductive ECNFs/AuNP	MG-63	No significant toxicityNo adverse effects on cell proliferation	[159]
Electrospun β-TCP/ECNFs membranes	PDLCs	 Biocompatible No adverse effect on cell affinity Degradable Improved cell growth 	[160]
Electrospun β-TCP@ECNFs membranes	MG-63	Cell proliferation on the membranesBiocompatible	[161]
Thermal-based electrospun ECNF/BG	MC3T3-E1	• Enhanced biomineralization, cell proliferation, and osteogenic differentiation	[162]
Sol-gel based Electrospun CNF/BG	MC3T3-E1	Improved formation of apatiteImproved cell proliferation	[152]
Electrospun CNF/BG	Rat BM-MSCs	• Enhanced biomineralization, cell adhesion and proliferation, and osteogenic induction	[163]
BC nanofibers-derived 3-D CNF/HAp	-	• Enhanced mineralization (HAp nucleation and growth) by HNO ₃ -surface treated CNFs	[156]
Electrospun ECNF/HAp	-	• Increased mineralization activity (the presence of apatite-like materials)	[164]
Sol-gel based ECNF-Si/Ca	Normal human skin fibroblasts NHOst	 Biocompatible Low cytotoxicity High ALP activity 	[165]
PP/CNF-HANR	Saos-2	 Improved mechanical strength (with Young's modulus from 1,360±20 to 2,517±15 MPa and tensile strength from 30.0±0.2 to 33.0±0.3 MPa for PP/2%/CNF–20% HANR) Promoted osteoblastic adhesion and viability on PP Biocompatible 	[166]

Table 3. Summary of carbon-based scaffolds for bone tissue engineering

Abbreviations: PLGA: poly(lactic-co-glycolic acid; HA/HAp : Hydroxyapatite; DCF: Direct current field; SNPs: Silica nanoparticles; Fe2O3: Ferrous sulfate; AuNP: Gold nanoparticle; β-TCP: β-tricalcium phosphate; BG: Bioglass; BC: Bacterial cellulose; ECNF-Si/Ca: Electrospun CNF-silica/calcium; PP/CNF–HANR: Polypropylene/carbon nanofiber-hydroxyapatite nanorod; PLA: poly(lactic acid); PCL: poly-ε-caprolactone; g-C3N4: Graphitic carbon nitride; MWCNTs: Multiwalled carbon nanotubes; f-MWCNT/Chitosan/B-GP: Functionalized multiwalled carbon nanotubes/chitosan/β-Glycerophosphate; MG-63: Human osteoblastic cell line; PDLCs: Human periodontal ligament cells; MC3T3-E1: Mouse osteoblastic cell line; BM-MSCs: Bone marrow-derived primary MSCs; NHOst: Normal human osteoblasts; Saos-2: Human osteoblast cell line; HNO₃: Nitric acid; ALP: Alkaline phosphatase.

3.2.3. Neural tissue engineering

TE could provide a promising strategy for the repair of the injured central nervous system (CNS) and peripheral nervous system (PNS). In this way, it is possible to treat neurodegenerative diseases such as Alzheimer's and Parkinson's diseases, traumatic brain injury (TBI), spinal cord injury (SCI), and peripheral nerve injury (PNI) by promoting neuronal growth [167]. PNI is a health problem mainly related to trauma characterized by serious loss of sensory, or motor control in affected patients that may lead to lifelong disability and disturbing quality of life [79, 168]. The self-regeneration and repair of damaged neural cells of peripheral nerves can be achieved if the nerve gap is less than 5 mm (by re-extension of the axon over the gap) [169], but greater defects may limit self-regeneration potential [170]. Hence, it is beneficial to utilize appropriate biomaterials as scaffolds in order to provide a 3D cell growth microenvironment, regenerate and treat the damaged nerves [72]. An ideal scaffold for NTE must exhibit an appropriate biomimetic structure, including physiochemical, mechanical, and electrical features along with being biocompatible, porous, and biodegradable [169, 171]. It should be noted that it is essential to employ scaffolds without any or fewer neuroinflammatory effects [172].

It is practical to use neurotrophic factors in company with scaffolds. As an example, Liu et al. [173] developed the electrospun aligned scaffolds based on regenerated silk fibroin (RSF) that were loaded with brain-derived neurotrophic factor (BDNF) and VEGF (dual-factor). They concluded that the resulting scaffolds could promote nerve regeneration for damaged peripheral nerve *in vitro* and *in vivo* studies. Other neurotrophic factors that could play essential roles in neural tissue regeneration include nerve growth factor (NGF), nerve growth factor (NT-3), glial cell line-derived neurotrophic factor (GDNF), and acidic and basic fibroblast growth factor (aFGF, bFGF) [174].

Other synthetic materials to fabricate neural tissue scaffolds are based on CNMs, including graphene, CNT, and CNFs [175]. Due to their natural conductivity features that are necessary for neural action potential, graphene and CNT have been chiefly utilized in NTE [176]. Nevertheless, the results of a previous study showed that polycarbonate urethane/VGCNFs composite as neural implant could decrease glial scar formation by limiting astrocyte functions. This substrate had positive interactions with neurons as well [177]. Another recent study describes that using electrospun PCL/VGCNFs as NGC was suitable for PC12 cell (rat pheochromocytoma cells) attachment and proliferation [178]. In light of the advantages of the electrospinning technique, its exploitation in fabricating nanofibrous scaffolds has been designed for NTE to act as both regenerative potential and a vehicle for therapeutic delivery to a neural lesion [79]. Mirzaei et al. [179] fabricated ECNFs and evaluated the effects of ECNFs alignment on the morphology and behavior of neural cells. They observed that the ECNFs alignment induced neural biomarkers and changed the morphology of cells from spherical to stretched cells aligned with the CNFs axis. **Table 4** Summarizes major carbon-based nanofibrous scaffolds applied in NTE applications.

 Table 4. Summary of carbon-based scaffolds for neural tissue engineering

Scaffold composition	Cells (in vitro)	Key findings
PCU:CN	Rat astrocytes	Minimized astrocyte functions on CNFs Decreased glipt score formation
		 Positive interactions with neurons
Electrospun PCL/CNF (conduit)	PC12 cells	• Suitable for cell attachment and proliferation
Electrospun CNFs (random	hEnSCs	• Biocompatible
and aligned topographies)		• CNF topography-dependent proliferation, morphology, and differentiation of hEnSCs:
		• Aligned CNFs: enhanced differentiation of hEnSCs into neurons and directed cell g
		axis, significant upregulation of neuronal markers (1.7- and 2.6- fold increase
		respectively in comparison with random CNFs), and downregulation of neural prog
		• Random CNFs: maintained progenitor state and slightly differentiation into c
		expression of nestin (3-fold increase) and slight upregulation of oligodendrocy
		comparison with aligned CNFs

Abbreviations: PCU:CN: Polycarbonate urethane: carbon nanofiber; PCL: Poly(ε-caprolactone); SWCNT/PLLA: Single-walled carbon-nanotube/Poly(L-I Polylactide-co-glycolide/carbon nanotube-dopamine; CS-PU/fMWCNT: Chitosan-polyurethane/functionalized multiwall carbon nanotubes; PPy: polypyrrole; rGONR: reduced graphene oxide nanoribbon; SiO2: Silicon dioxide; rGONMs: reduced graphene oxide nanomeshes; GOF: Graphene oxide foam; PC12: Rat phe Human endometrial stem cells; OEC: Olfactory ensheathing glial cells; S42: Schwann cells; L929: Mouse fibroblast cell line; SCs; Schwann cells; BV2: Muri human neural stem cells; NE-4C: Neuroectodermal stem cells.

3.2.4. Musculoskeletal tissue engineering

The musculoskeletal system contains bones (the skeleton), muscles, cartilage, tendons, ligaments, and joints [180, 181]. BTE has been discussed in section 3.2.2. Many tissues of an adult human body, such as cardiac tissue, have limited self-regeneration potential after injuries. Hence, TE could be an appealing option to be employed in these conditions [182]. For myocardial TE, several studies have also utilized CNMs. Stout et al. [183] assessed the effects of PLGA-VGCNFs composites on human cardiomyocytes. This cardiovascular patch was cytocompatible and increased cardiomyocyte adhesion and proliferation in the dead tissue zone. Typically, incorporating VGCNFs into polymeric scaffolds, such as PLGA, improves or increases their mechanical conductivity, cytocompatibility, and other properties [181, 184]. Interestingly, along with electrospun CS-PVA-MWCNT scaffold, electrical stimulation and small molecules such as CHIR99021 (CHIR) have been applied to induce cardiac differentiation of unrestricted somatic

stem cells (USSCs) for cardiac TE purposes [185]. Electrical stimulation and conductive scaffolds (*e.g.*, CNF/CNT) are important factors for the attachment and growth of certain cells, especially skeletal muscle, cardiac, neural cells, and osteoblasts [128, 185]. Most studies on CNMs for musculoskeletal engineering are developed by utilizing graphene, and CNTs [186]. For meniscal TE, Gopinathan et al. [187] designed customizable functionalized 3D PCL scaffolds reinforced with VGCNFs. They showed its mechanical integrity, electrical properties, and cytocompatibility in primary meniscal cells.

4. Clinical and preclinical studies

To the best of our knowledge, CNMs, including CNFs, ECNFs, and CNTs, have not yet been used clinically, although a plethora of *in vitro* studies suggests these materials' TE potential. The important reason behind it is safety issues [188]. However, few *in vivo* studies (animal experiments) are being conducted based on *in vitro* findings. Herein, we mentioned certain *in vivo* studies. *In vivo* study for VGCNFs gauze was performed on rat back-bleeding models and promoted coagulation and reduced blood loss as a blood-repelling CNFs gauze. CNF gauze was applied to the rat skin for skin compatibility tests with no itching or erythema observed after 12 h [141]. New ectopic bone formation was reported in a study that a 3D thin electrospun carbon fiber web (TCFW) and recombinant human bone morphogenetic protein 2 (rhBMP-2) as a growth factor was implanted into the dorsal muscle of a ddY mouse (ddY stands for Deutschland, Denken, and Yoken). It showed good bone-tissue compatibility without necrosis or significant inflammatory responses. The rhBMP-2/TCFW composite was also implanted into the rat ileum with critical-sized defects and repaired the orthotopic defect [189]. In another similar study, the ECNF sheets functionalized by HA and bovine serum albumin (BSA) as a growth factor were biocompatible,

with fewer inflammatory cells implanted subcutaneously in a rabbit model. It is suggested that CNF/HA/BSA sheets as periosteum-like membranes have the potential for BTE [190].

In another recent survey, the osteoconductive effects (with the highest new bone formation (61.3 \pm 4.2%)) of ECNFs decorated with HA crystal (CNFs/HA) have been exhibited in a rat's femur defect [33]. Another study reported the *in vivo* biocompatibility of porous ultra-short single-walled carbon nanotubes (US-tubes)-reinforced poly(propylene fumarate) (PPF) nanocomposites in a New Zealand rabbit model. The nanocomposite scaffolds were implanted into subcutaneous pockets (in the dorsum of the rabbits), and femoral condyles (to analyze soft and hard tissue responses). High bone tissue ingrowth, reduced inflammatory cell density, and increased connective tissue organization were observed 12 weeks after implantation. They have suggested that US-tube/PPF scaffolds are osteoconductive, and bioactive assisting osteogenesis [191].

Farzamfar et al. [178] showed peripheral nerve regeneration using a conductive electrospun PCL/VGCNFs conduit as a neural guidance channel. This conduit exhibited no toxicity or immunogenic reactions after being implanted into a rat model of sciatic nerve defect. This study suggested that it could be a promising strategy for treating PNI in the clinic. *In vivo* study by Gopinathan et al. [187] for nanocomposite 3D PCL scaffold with VGCNFs in New Zealand white rabbits. In this study, subcutaneous implantation of nanocomposite was performed to evaluate it's *in vivo* biocompatibility and biotoxicity to be employed as a meniscus scaffold. Their study demonstrated that nanocomposite had no toxicity. Concerning the growing number of *in vitro* and *in vivo* works pertinent to CNMs and their findings, it is anticipated that these materials, in particular ECNFs hold promise to be employed in clinical studies in the future.

5. Concluding remarks and future perspective

Nanomaterials and nanotechnology have inevitable impacts on modern devices and procedures in various fields of biomedicine. Carbon-based nanomaterials have shown promising and fascinating performance in diagnosing, treating, and recovering different diseases. Among them, CNFs have attracted significant attention due to their outstanding physicochemical properties. It has been shown that the synthesis method of CNFs (CVD growth and carbonization) has critical effects on the physicochemical properties and, subsequently, the performance of CNFs in the intended application. Although the non-spinning-based synthesis methods (*e.g.*, CVD) provide CNFs with higher mechanical strength and electrical conductivity, they suffer from high synthesis costs and toxicological issues but suffer from lower graphitization, electrical properties, mechanical strength, and more structural defects. However, the structural defects can be exploited for biomolecules conjugation/attachment and is an advantage from the biomedical point of view.

A critical issue in applying CNFs for biomedical applications and the clinical translation, except for biosensors, is the biopersistence, biodegradation and the in vivo fate of CNFs. It has been reported that immune cells (e.g., neutrophils and macrophages) and some enzymes (e.g., peroxidases) can degrade carbon-based nanomaterials [192]. Moreover, the studies implied that the type of surface functional groups, structural defects, and oxygen percentage determine the biodegradation profile and biopersistence of CNFs [193]. In this concept, it can be assumed that the electrospun CNFs are more biodegradables than the CVD-based CNFs due to more structural defects and amorphous nature. However, comprehensive studies should be carried out using the big data-based experimental methods, such as structure-activity relationship (QSAR), artificial neural network (ANN), and design of experiment (DOE) approaches to precisely assess the biodegradation and biopersistence of each type of CNFs and the effect of the involved synthesis parameter and post-treatments. Nevertheless, the authors believe that CNFs have the potential to be considered innovative and enabling structures beneficial for various filed of biomedicine.

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The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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