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Development and evaluation of budesonide-based modified-release liquid oral dosage forms

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PII: S1773-2247(19)30732-4

DOI: <https://doi.org/10.1016/j.jddst.2019.101273>

Reference: JDDST 101273

To appear in: *Journal of Drug Delivery Science and Technology*

Received Date: 22 May 2019

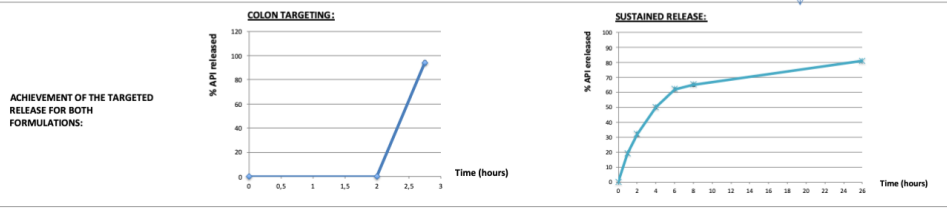
Revised Date: 31 August 2019

Accepted Date: 9 September 2019

Please cite this article as: F. Ronchi, A. Sereno, M. Paide, Ismaë. Hennia, P. Sacré, G. Guillaume, V. Stéphane, J. Goole, K. Amighi, Development and evaluation of budesonide-based modified-release liquid oral dosage forms, *Journal of Drug Delivery Science and Technology* (2019), doi: <https://doi.org/10.1016/j.jddst.2019.101273>.

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23 Keywords: Budesonide, multi-layered particles, liquid syrup, oral dosage form, colon targeting, sustained  
24 release.

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**31 Table of abbreviations and acronyms**

32 API: Active pharmaceutical ingredient

33 BUD: Budesonide

34 CV: Coefficient of variation

35 GIT: Gastrointestinal tract

36 GMS: Glyceryl monostearate

37 HCL: Hydrochloric acid

38 HPLC: High performance liquid chromatography

39 K30: Kollidon<sup>®</sup> 30

40 LOQ: Limit of quantification

41 MUDF: Multiple-unit dosage form

42 MUPS: Multi-unit pellets system

43 PBS: Phosphate-buffered saline

44 PVP: Polyvinylpyrrolidone (Povidone)

45 RH: Relative humidity

46 rpm: Rotation per minute

47 SEM: Scanning electron microscopy

48 SOP: Standard operating procedure

49 SUDF: Single-unit dosage form

50 T°: Temperature (°C)

51 TEC: Triethyl citrate

52 TGA: Thermogravimetric analysis

53 UK: United Kingdom

54 USA: United States of America

55

56 **Abstract**

57 Modified-release oral drug delivery dosage forms are widely used in the pharmaceutical field to  
58 overcome all the potential issues imposed by the physiological variabilities of the gastrointestinal  
59 tract as well as to maintain drug concentrations within the therapeutic window. In the market, they  
60 are available only as solid dosage forms such as capsules or tablets. The development of a liquid  
61 oral dosage form with modified-release properties has been keenly awaited. This form could  
62 increase the compliance of patients with a swallowing impairment (i.e. paediatric, older or critically  
63 ill patients) and, consequently, the efficacy of the therapeutic treatment. In this study, budesonide  
64 was used as a model drug to develop a modified-release liquid oral dosage form (i.e. colonic-  
65 release, sustained-release). For this purpose, multi-layered particles were obtained, starting from  
66 small microcrystalline cellulose neutral cores (Cellets<sup>®</sup> with a mean diameter lower than 500  $\mu\text{m}$ ),  
67 in a lab-scale fluid-bed coater. Poly(meth)acrylate polymers commonly available under the trade  
68 name of Eudragit<sup>®</sup>, such as S100, RS PO, RL100 and E100, were used to get defined drug release  
69 profiles. They were also used to guarantee the stability of the reconstituted liquid syrup during 2  
70 weeks of storage at room temperature.

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## 80 **1. Introduction**

81 Among the different routes of administration, the oral route is still the most commonly used due to  
82 its ease of administration [1]. Oral dosage forms can be classified into immediate- and modified-  
83 release systems. Although immediate-release oral dosage forms may provide a rapid onset of  
84 therapeutic response, they are not able to control the release of a drug [2]. Such dosage forms have  
85 to be administered several times per day to maintain the drug concentration within the  
86 therapeutically effective range for the whole duration of the treatment. This can result in fluctuating  
87 drug levels in the bloodstream and in a decrease in compliance by patients [3, 4]. Currently,  
88 modified-release formulations, such as sustained-release systems, are being used to treat chronic  
89 illness [4, 5]. Advantages of modified-release dosage forms include their ability to provide temporal  
90 or spatial control of the release thanks to the use of suitable polymeric excipients [6, 7]. For  
91 bioavailability, efficacy or safety reasons, the selective delivery of drugs to specific gastrointestinal  
92 sites is pursued. This is done to protect the loaded drugs from biological fluids or potential issues  
93 imposed by the physiologic variabilities of gastrointestinal tract (GIT) (e.g. pH, the commensal  
94 flora, enzymatic activity, surface area and gastrointestinal transit time). It also prevents delivery of  
95 the drug outside the so-called absorption window, ensuring its release in the target site as  
96 extensively as possible [8]. An example of this latter case is colon delivery, for which it is necessary  
97 to prevent the release of active ingredients in the entire gastric residence and the small intestinal  
98 transit. Site-selective release is to be sought based on environmental differences between the small  
99 and the large intestine, such as the quali-quantitative composition of the microbiota, the pH of  
100 fluids, the intraluminal pressure and the transit time [6, 9, 10].

101 Solid dosage forms are the most frequently used drug delivery systems for oral administration [7].  
102 They differ in both size and the number of units administered as a single dose, including single-unit  
103 dosage forms (SUDF) and multiple-unit dosage forms (MUDF) [11]. The most important  
104 characteristic of MUDFs in comparison to SUDFs is their lower susceptibility to dose dumping and  
105 to a faster gastric emptying. This is because the subunits of a MUDF may be distributed more

106 evenly throughout the gastro intestinal tract [12, 13]. Such advantages result in fewer adverse  
107 effects, better bioavailability, lower variability in the drug release and, consequently, better  
108 compliance by patients [14].

109 However, the ability to take such solid dosage forms may be compromised in patients with  
110 swallowing impairments (dysphagia), especially paediatric, elderly or critically ill patients [15-18].  
111 Swallowing issues have been described as dosage forms getting stuck in the throat, an  
112 uncomfortable feeling, the need for repeated swallowing attempts, gagging, choking, coughing  
113 while swallowing or vomiting [19]. It may result in alteration of the dosage form, omission of doses  
114 or discontinuation of medications [20, 21, 16]. Therefore, a modified-release liquid oral dosage  
115 form, has been highly awaited for patients with swallowing impairment.

116 In a previous study, a new technology based on gastro-resistant small omeprazole-loaded multi-  
117 layered pellets dispersed in syrup was developed [22]. In this study, budesonide (BUD), a potent  
118 corticosteroid used in the management of asthma and allergic rhinitis, the treatment of various skin  
119 disorders and the treatment of inflammatory bowel diseases (e.g. ulcerative colitis, Crohn's  
120 disease), was used as a model drug [23-26]. BUD is commercially available in the form of pH-  
121 dependent enteric-coated preparations, mainly for the local treatment of Crohn's disease, a chronic  
122 inflammatory bowel disorder of unknown aetiology that may affect any part of the gastrointestinal  
123 tract in both children and adults. However, the most common sites of inflammation are the distal  
124 ileum and/or the ascending colon [27]. BUD was employed to verify the feasibility of this new  
125 technology. The technology was constituted of multi-layered particles dispersed in a liquid vehicle,  
126 with diverse kinds of release (i.e. colon-targeting and sustained-release).

127 The aim of this work was to demonstrate the feasibility of the new technology by developing  
128 budesonide-based modified-release liquid oral dosage forms that present a delayed release of the  
129 drug in the colon or a sustained release, and to verify the short-term stability of drug release after  
130 dispersion of the multicoated particles in a syrup.

131

## 132 **2. Materials and methods**

### 133 *2.1 Materials*

134 Microcrystalline cellulose pellets (Cellets<sup>®</sup> 263, Process Center GMBH & Co, Germany) with a  
135 mean diameter, D(50), ranged between 200 and 300 µm were used as neutral core. Budesonide  
136 (Sterling Spa, Corciano, Italy) was used as a model drug. Eudragit<sup>®</sup> S100 (an anionic copolymer  
137 based on methacrylic acid and methyl methacrylate (1:2)), in the form of powder (Evonik<sup>®</sup>  
138 Industries, Darmstadt, Germany), was used as a colon-targeting polymer. Eudragits RS PO and RL  
139 100 (insoluble copolymers of ethyl acrylate, methyl methacrylate and a low content of a  
140 methacrylic acid ester with quaternary ammonium groups (1:2:0.1 for RS PO and 1:2:0.2 for RL  
141 100, respectively)) in the form of powder and granules (Evonik<sup>®</sup> Industries, Darmstadt, Germany)  
142 were used to provide sustained-release formulations. Eudragit<sup>®</sup> E100 (cationic copolymer based on  
143 dimethylaminoethyl methacrylate, butyl methacrylate and methyl methacrylate in the ratio 1:2:1) in  
144 the form of granules (Evonik<sup>®</sup> Industries, Darmstadt, Germany) was used as a gastro-soluble  
145 polymer. Povidone (Kollidon<sup>®</sup> K30, D-BASF, Germany) and talc (micronized 10 µm talc, Sigma  
146 Aldrich, USA) were used as a binder and a bulk agent, respectively. Microfine lactose (Lactochem<sup>®</sup>  
147 Microfine 201, Borculo Domo, Netherlands) was used as hydrophilic agent. Triethyl citrate (TEC,  
148 Alfa Aeser, USA) was used as plasticizer. Glyceryl monostearate (GMS, D-BASF, Germany) and  
149 aluminium oxide (Sigma Aldrich, United States) were employed as bulk agents because of their  
150 anti-electrostatic properties.

151 Neosorb sorbitol (Roquette, France), Avicel<sup>®</sup> RC-591 (microcrystalline cellulose and sodium  
152 carboxymethylcellulose, FMC, USA), Kollidon<sup>®</sup> K30 (polyvinylpyrrolidone, PVP, D-BASF,  
153 Germany), anhydrous sodium carbonate (Sigma Aldrich, United States) and dihydrate disodium  
154 hydrogen phosphate (Merck, Germany) were used for the preparation of the dry syrup.

### 155 *2.2 Production of multicoated particles*



156 Neutral microcrystalline cellulose pellets (1-1.5 Kg) were transferred into a lab-scale bottom-spray  
157 fluidized bed coater (SLFLL\_3, Lleal s.a., Spain) fitted with a Würster insert.

158 To provide a colonic release, the pellets were coated with four successive coating layers: a drug  
159 layer; a colon-targeting polymer layer to prevent the early release of budesonide in the upper part of  
160 the GIT; an isolating layer to avoid interaction between layers 2 and 4; and a gastro-soluble  
161 polymer layer to avoid drug release in the syrup. To provide a sustained release, the pellets were  
162 coated with three successive coating layers: a drug layer; a sustained-release layer including two  
163 insoluble polymers in a ratio to obtain a progressive release of budesonide over a prolonged period  
164 of time; and a gastro-soluble polymer layer to avoid drug release into the syrup. The compositions  
165 of the coating dispersions are presented in the section “Results and discussions”.

166 For both formulations, all the coating dispersions were filtered through a 200 µm sieve before  
167 starting the coating processes. During all the coating processes, the dispersions were continuously  
168 stirred to prevent sedimentation of insoluble particles. The coated pellets obtained after each  
169 coating steps (1<sup>st</sup>, 2<sup>nd</sup>, 4<sup>th</sup> for the colon-targeted formulation and 1<sup>st</sup>, 2<sup>nd</sup>, 3<sup>rd</sup> for the sustained-release  
170 formulation) were sieved to discard potential agglomerates before continuing the coating process  
171 with the next coating layer. After the sieving (30 Hz for 5 min, 500 µm sieve, Rhewum vibrating  
172 apparatus, Germany) only 1 Kg of the coated pellets obtained from the previous step was used for  
173 the next coating step.

#### 174 *2.2.1 Preparation of coating dispersions for colon targeting*

175 For the preparation of the first coating dispersion, budesonide and PVP were dissolved in  
176 ethanol using a gentle stirring system (blade stirrer, Janke & Kunkel, model RW20, Ika  
177 Labortechnik, Germany). Then, talc was dispersed at 300 rpm. The second coating dispersion  
178 contained Eudragit<sup>®</sup> S100, which was solubilized in an isopropanol-water mixture (87:13) at 500  
179 rpm for 2 hours using the blade stirrer. TEC and talc were then added under the same conditions.

180 The third coating suspension was an ethanolic solution composed of PVP, in which talc was  
 181 suspended at 300 rpm using the blade stirrer.

182 The outermost layer dispersion was prepared by solubilizing the Eudragit<sup>®</sup> E100 granules in an  
 183 ethanol-water mixture (6:94) at 500 rpm. After its complete solubilization, GMS and aluminium  
 184 oxide were dispersed using a T25 Ultra-Turrax (IKA<sup>®</sup>, Staufen, Germany) at 13 500 rpm. Then, talc  
 185 was added to the suspension at 500 rpm using the blade stirrer.

186 The coating parameters used for the application of each coating layer in the fluid-bed coater are  
 187 listed in Table 1.

188

189 Table 1. Coating parameters used to produce the budesonide colon-targeted formulation

Coating parameters	First coating: drug layer	Second coating: delayed-release layer	Third coating: isolating layer	Fourth coating: protective coating layer
Inlet air temperature (°C)	35-37	40-42	33-35	33-35
Outlet air temperature (°C)	28-30	32-34	26-28	26-28
Product temperature (°C)	30-32	35-37	28-30	28-30
Air flow (m <sup>3</sup> /h)	15-25	90-95	90-95	70-85
Spraying rate (g/min)	10-12	10-12	10-12	10-12
Air pressure (Bar)	1.5-1.7	1.7-1.9	1.6-1.8	1.6-1.8

190

191

### 192 2.2.2 Preparation of coating dispersions for sustained release

193 As usual, for the preparation of the first coating dispersion, the soluble compounds were previously  
 194 dissolved in ethanol-budesonide and PVP with a blade stirrer (Janke & Kunkel, model RW20, Ika

195 Labortechnik, Germany). Microfine lactose was dispersed using a T25 Ultra-Turrax (IKA<sup>®</sup>,  
 196 Staufen, Germany) at 13.500 rpm. Then, talc was added using the blade stirrer at 300 rpm.  
 197 The second coating dispersion contained Eudragit<sup>®</sup> RS PO and RL100, which were dissolved in an  
 198 isopropanol-water mixture (with the ratio 87:13) at 500 rpm for 2 hours. Then, TEC and talc were  
 199 dissolved and dispersed, respectively. Considering the nature of polymers employed in the second  
 200 coating layer, this approach prevents the potential risk of interaction with Eudragit<sup>®</sup> E100 as all the  
 201 polymers used are positively charged. In this case, the use of isolating layers was not necessary. The  
 202 last coating dispersion was equivalent to that used for the colon-targeting formulation. For this  
 203 reason, the preparation was made following the same procedure.  
 204 The coating parameters used for the application of each coating layer are listed in Table 2.

205  
 206 Table 2. Coating parameters used to produce the budesonide sustained-release formulation

Coating parameters	First coating: drug layer	Second coating: sustained-release layer	Third coating: protective coating layer
Inlet air temperature (°C)	33-35	28-30	33-35
Outlet air temperature (°C)	25-27	25-27	26-28
Product temperature (°C)	26-28	26-28	28-30
Air flow (m <sup>3</sup> /h)	15-25	70-85	70-85
Spraying rate (g/min)	10-12	10-12	10-12
Air pressure (Bar)	1.5-1.7	1.7-1.9	1.6-1.8

207

## 208 2.3 Characterization of multicoated pellets

### 209 2.3.1 Coated-pellet agglomerates

210 To determine the amount of coated-pellet agglomerates and to discard any potential agglomerates,  
211 the coated pellets obtained after each step were sieved (i.e. after the 1<sup>st</sup>, 2<sup>nd</sup> and 4<sup>th</sup> coating for the  
212 colon-targeted formulation and the 1<sup>st</sup>, 2<sup>nd</sup>, and 3<sup>rd</sup> coating for the sustained-release formulation).  
213 This step avoids multiple coating or the presence of broken agglomerates during the next steps in  
214 the coating process. The sieving was done at 30 Hz for 5 min, using a 500 µm sieve (Rhewum  
215 vibrating apparatus, Germany). From the sieved pellets obtained from the previous step, 1 Kg was  
216 then used as a starting point for the next coating step.

217

### 218 *2.3.2 Particle-size distribution*

219 For both formulations, the particle-size distribution of the multicoated pellets was evaluated after  
220 each coating step, after withdrawing the agglomerates by sieving. This evaluation was made by  
221 laser diffraction (Mastersizer<sup>®</sup> 3000, Malvern Instruments, UK), using a dry sample dispersion  
222 accessory (Aero S). The SOP (Standard operating procedure) used for the analysis was: Fraunhofer  
223 scattering; dispersive air pressure, 0.5 bar; vibrating rate, 50%; measurement time, 10s.

224

### 225 *2.3.3 Quantification of budesonide*

226 For both formulations, the drug content was determined by grinding the coated pellets in a mortar  
227 and weighing an amount of powder containing a theoretical content equivalent to 3 mg of  
228 budesonide. The drug was extracted using an ethanol-water-methanol mixture in the ratio  
229 7:37.2:55.8 under sonication.

230 The high performance liquid chromatography (HPLC) system used was a series 1200 Agilent  
231 Technologies system (USA), equipped with a single pump, an autosampler and a diode-array UV  
232 detector. The column used was an end-capped RP-18 (Purospher<sup>®</sup> STAR, Germany). The  
233 chromatographic conditions were set as follows: wavelength, 225 nm; flow rate, 1.5 mL/min;  
234 temperature, 30°C; injection volume, 100 µL; run time, 8 min. The LOQ (limit of quantification) of

235 this method coincided with 0.2  $\mu\text{g}/\text{mL}$  and the LOD (limit of detection) coincided with 0.5  $\mu\text{g}/\text{mL}$ ,  
236 which corresponded to 0.01% and 0.02% w/w budesonide content in the coated dry  
237 pellets, respectively.

238 All the data are the means of five determinations. The coating process efficiency was determined by  
239 expressing the mean drug content as a percentage of the theoretical drug loading.

240

#### 241 2.3.4 *Thermogravimetric analysis*

242 To evaluate the residual amount of solvent after each coating layer and in the final batch,  
243 thermogravimetric analysis (TGA) (Q500, TA Instruments, USA) was performed. Approximately  
244 10 mg of intact coated sample was loaded in platinum pans and heated from 30°C to 170°C at a  
245 heating rate set at 10°C/min.

246

#### 247 2.3.5 *Characterization of the film structure*

248 The morphology of both the external surfaces and inner structure of the multicoated pellets was  
249 evaluated using scanning electron microscopy (SEM). For the analysis of the external surface, a  
250 sample of several pellets were fixed and “sprinkled” onto a conductive, adhesive tape placed on a  
251 sample holder. All preparations were sputtered with gold/palladium to obtain an electrically  
252 conductive surface. The samples were immediately analysed to avoid any change in characteristics.  
253 The analysis was performed using a high-resolution field emission scanning electron microscope (7  
254 kV, Jeol JSM using an oil diffusion pump 6610LA, Tokyo, Japan). For the cross-section analysis,  
255 pellets were embedded in a light curing resin (Heraeus Technovit 2200 light cure). After hardening  
256 ( $\approx$  2 minutes), the embedding was cooled in liquid nitrogen. Then, the embedding was fractured  
257 using a plier and dried before sputtering with gold/ palladium to obtain an electrically conductive  
258 surface (10 kV, Jeol JSM using a turbo molecular pump IT 300, Tokyo, Japan). The coated pellets

259 were observed at different magnifications between 250x, 500x and 1000x.

260

### 261 2.3.6 Dissolution test

262 A Distek 2100C USP 29 dissolution apparatus (Distek Inc., North Brunswick, NJ, USA), Type II  
263 (paddle method), with a rotational speed of 100 rpm and a temperature set at 37°C, was used for the  
264 dissolution tests.

265 For the colonic-release formulation, the dissolution test was carried out for 2 hours in 750 mL of  
266 acid medium pH 1.2 (HCL 0.1N) and for 45 minutes in 1 000 mL of phosphate buffer medium  
267 (PBS) [0.05M] at pH 7.5. Indeed, to mimic distal intestine pH value (pH 7.5), 250 mL of tri-sodium  
268 phosphate dodecahydrate was added to the acidic medium. The dissolution tests on the sustained-  
269 release budesonide multi-layered pellets were performed in one of two different manners,  
270 depending on the coating step at which the test was done. When the dissolution test was performed  
271 on the pellets obtained after the modified-release coating (2<sup>nd</sup> coating), 1 000 mL of PBS 0.05 M at  
272 pH 7.5 was employed as a buffer medium and 6 mL of sample was withdrawn after 1, 2, 4, 6, 8 and  
273 24 hours. The withdrawn volumes were replaced with equal volumes of blank medium to maintain  
274 the volume of the dissolution medium constant during the whole test. When the dissolution test was  
275 performed after the protective-coating layer (3<sup>rd</sup> coating), the test involved 750 mL of acid medium  
276 pH 1.2 (HCL 0.1N) for 2 hours followed by 22 hours in PBS 0.05 M at pH 7.5. As before, 6 mL of  
277 sample was withdrawn after 1 and 2 hours in the acidic medium, and 1, 2, 4, 6, 8 and 22 hours in  
278 the buffer medium and replaced with an equal volume of blank medium. A maximum dissolution  
279 test duration of 24 hours was selected because when using a liquid dosage form based on a  
280 suspension of small particles, GI transit times are shorter than for solid dosage forms.

281 The amount of drug released was detected by HPLC (Agilent, USA) in both acid and phosphate  
282 buffer medium (see 2.3.3. Quantification of budesonide) after filtration and propriate dilution of the  
283 withdrawn samples. The dissolution tests were performed on dry multi-layered pellets and on multi-  
284 layered pellets dispersed extemporaneously in the reconstituted syrup; in both cases, the syrup

285 contained 3 mg of budesonide. The percentages of drug released were quantified at the  
286 predetermined times and averaged (n=6).

287  
288  
289 *2.4 Preparation of the syrup and stability of the dispersed pellets*

290 To evaluate the stability of the budesonide-loaded multicoated pellets in a liquid dosage form, a  
291 “conventional” syrup was prepared. Kollidon<sup>®</sup> 30 (10% w/w), sorbitol (60% w/w) and the buffering  
292 agents (disodium hydrogen phosphate dihydrate) were dissolved in water at  $60^{\circ}\text{C} \pm 5^{\circ}\text{C}$  at 500 rpm  
293 using a blade stirrer (Janke & Kunkel, model RW20, Ika Labortechnik). Then, Avicel<sup>®</sup> RC-591 (2%  
294 w/w) was dispersed at 400 rpm. The suspension was cooled at room temperature and the final  
295 volume was adjusted with water to 100 mL. The pH was set at  $7.5 \pm 0.2$  with sodium carbonate  
296 (0.24% w/w). For both formulations, multi-coated pellets were added to reach a concentration of  
297 budesonide of 3 mg/10 mL of dispensed dose and the syrups were poured into closed amber glass  
298 bottles. The final preparations were stored at  $25^{\circ}\text{C}/50\% \pm 5 \text{ RH}$ .

299 Both the drug content and dissolution profiles of budesonide were evaluated at time zero, and after  
300 1 and 2 weeks of storage using the methods described before. To quantify the amount of budesonide  
301 in the syrup (i.e. inside and outside the pellets), 10 mL of sample was dispersed in a mixture made  
302 of 20 mL purified water and 40 mL ethanol before being sonicated for 20 minutes. Then, another 30  
303 mL of ethanol was transferred into a 100 mL amber volumetric flask to be sonicated for other 30  
304 minutes. Samples obtained from an appropriate dilution (1:10) of the filtered extracts with the  
305 dilution phase (methanol-water 60:40) were injected into the HPLC to be quantified.

306 The amount of budesonide that was diffused outside the pellets in the syrup during the storage was  
307 also investigated. A sample of 5 mL of the syrup was dispersed in 10 mL of ethanol before being  
308 centrifugated for 5 minutes, at 2 000 rpm and at  $20^{\circ}\text{C}$  (Haerus Multifuge X1R centrifuge, Thermo  
309 Scientific, USA). Samples obtained from an appropriate dilution (1:150) of the filtered extracts with  
310 the dilution phase (methanol-water 60:40) were injected into the HPLC to be quantified.

311

312 *2.4.1 Statistical evaluation*

313 The Student's t-test was used to determine whether the means of two sets of data were significantly  
314 different from each other. In this case, the test was done on the amount of BUD released at time  
315 zero from the colon-targeted formulation, compared with the amount obtained after 1 and 2 weeks  
316 of storage. If the  $p$  value is higher than 0.05, no significant differences are present between the two  
317 sets of data [28].

318 In addition, the similarity factor ( $f_2$ ) was used to determine the similarity of dissolution profiles for  
319 the sustained-release formulation. This test is recommended in the FDA's Guidance for Industry for  
320 profiles that include different time points, such as this one. To compare the dissolution profiles  
321 obtained, the same test conditions were set and the same dissolution time points were examined. If  
322 the  $f_2$  value is included in the range between 50 and 100, the two dissolution profiles are considered  
323 similar [29].

324

325

326

327 **3. Results and discussion**

328

329 *3.1 Characterization of multicoated particles*330 *3.1.1 Optimization of the coating procedure*

331 The lab-scale bottom-spray fluid-bed coater used to produce both budesonide formulations is a  
332 prototype adapted for coating small particles with polymeric films [22]. It has the following  
333 structural components: a hole distribution plate composed of eight concentric circumferences with  
334 holes of different diameters, 4.0 mm, 0.8 mm and 1.8 mm, in sequence respectively; a main  
335 chamber 96 cm high; and a metal filter structure with a size aperture of 250  $\mu\text{m}$ . Also, the position  
336 of the cylinder (placed 3 cm away from the perforated plate) was adjusted to improve the particle  
337 flow and the distribution of coating on the particle surfaces.



338

339 *Budesonide-loaded colonic-release multicoated pellets*

340 During the first coating, the main issue was shown to be the spray drying of the ethanolic dispersion  
341 when the inlet temperature was too high. It was demonstrated that the yield of the process was  
342 improved from 80% to 95% when the inlet temperature was set and limited to a 35-37°C range,  
343 with a corresponding product temperature of 30-32°C.

344 The pH in the caecum and in the ascending colon drops to slightly acidic values because of an  
345 anaerobic bacterial metabolism that results in a local accumulation of short-chain fatty acids.  
346 However, the traverse and descending branches are restored to a neutral to slightly alkaline  
347 environment due to the absorption of fermentation products. In vivo pH variabilities in the intestine  
348 do not permit a specific drug release to be predicted at the entry to the colon. However, such  
349 changes in the pH have been studied to achieve colon-targeting of drugs through the application of  
350 pH-sensitive coatings [30, 31]. To avoid an early release of drug in the upper part of the  
351 gastrointestinal tract and to attempt to release it at the correct site of the colon, Eudragit<sup>®</sup> S100 was  
352 used in the second coating as the pH-sensitive colonic polymer. Different percentages of coating  
353 were evaluated from 17% to 25% w/w of total solids, which corresponded to 12.4% to 18.1% w/w  
354 of Eudragit S<sup>®</sup>100. The aim was to evaluate the appropriate amount of coating to allow a good  
355 protection in acidic gastric medium as well as a fast drug release in intestinal buffer medium at  
356 around pH 7.5, which corresponds to that of the ileum (i.e. the last section of the small intestine  
357 before entering to the ascending colon). The results obtained in this paper demonstrate that the  
358 formulation selected for the second coating was able to provide gastro-resistance and a high release  
359 of budesonide in the buffer medium at pH 7.5 during the dissolution test (see 3.4).

360 The third coating was an ethanolic dispersion of PVP and talc. This coating aimed to avoid potential  
361 interaction between the anionic and the cationic polymers present in the second and in the fourth  
362 coating, respectively.

363 For the fourth coating, the main issue was the difficulty in reaching sufficient integrity of the  
 364 protective film based on Eudragit E<sup>®</sup>100, a cationic polymer. The addition of hydrophobic  
 365 substances to the polymer was able to increase the strength of the protective film as a barrier against  
 366 the liquid vehicle in the reconstituted syrup. This barrier prevented, or at least decreased, the  
 367 liquid's swelling and gel-forming properties, which are normally activated after prolonged contact  
 368 with neutral pH media [32]. The composition of the selected formulation is shown in Table 3.

369

370 Table 3: Qualitative and quantitative compositions (%w/w) of each layer present in the selected budesonide-loaded  
 371 colonic-release multi-layered pellets

Substance	Core	First coating: drug layer	Second coating: delayed-release layer	Third coating: isolating layer	Fourth coating: protective coating layer
Neutral core	48.56				
Budesonide		0.73			
PVP		0.24		2.01	
Talc		0.24	3.04	2.01	6.57
Eudragit S100			12.44		
TEC			1.69		
Eudragit E100					20.45
GMS					1.01
Aluminium oxide					1.01
Ethanol		√		√	√
Isopropanol			√		
Water			√*		√*
<b>% coating</b>		<b>1.21</b>	<b>17.17</b>	<b>4.02</b>	<b>29.04</b>

372 \* The percentages of water in the coating dispersions were 15% and 6% for the second and fourth coating layers,  
 373 respectively.

374

### 375 *Budesonide-loaded sustained-release multicoated pellets*

376 In addition to the spray-drying issue already encountered with the colonic-release multi-layered  
 377 pellets, the poor solubility of budesonide (BUD) in water (10.7 mg/ L at 25°C) was another issue to  
 378 solve when developing a sustained-release dosage form. Therefore, it was decided to add lactose as  
 379 a hydrophilic and hydrosoluble compound to increase the hydrophilic environment around the drug  
 380 upon contact with aqueous fluids. As ethanol was used as a solvent to dissolve budesonide and PVP

381 for the first coating, micronized lactose and talc were used to obtain a homogeneous coating  
382 suspension.

383 The second coating was made of water-insoluble polymers to obtain a sustained release of BUD.  
384 Eudragit<sup>®</sup> RS PO and RL 100 were selected as they were compatible with each other, allowing the  
385 use of different ratios to modulate the release profiles of BUD from the developed sustained-release  
386 multilayered pellets. Two batches produced with 100% w/w of each polymer were used as  
387 references for high (RL) and low (RS) permeability barrier films to control the drug release. Three  
388 polymer ratios were also evaluated, namely RS-RL 50:50, RS-RL 70:30, and RS-RL 30:70 to  
389 obtain coatings with intermediate permeabilities between the low- and high- permeability barrier  
390 films. The results obtained in this paper demonstrate that the formulation selected for the second  
391 coating was able to provide a sustained release of BUD during the 24-hour dissolution test done in a  
392 PBS 0.05 M buffer medium pH 7.5.

393 After the second coating, the outermost coating was based on Eudragit<sup>®</sup> E100. This coating was  
394 similar to that used for the colon-targeted coated pellets and was applied to avoid the early release  
395 of the drug in the reconstituted syrup during storage. The composition of the selected formulation is  
396 shown in Table 4.

397 Table 4: Qualitative and quantitative compositions (% w/w) of the selected formulation for budesonide sustained-release  
398 pellets

Substance	Core	First coating: drug layer	Second Coating: sustained-release layer	Third coating: protective coating layer
Neutral core	57.83			
Budesonide		0.87		
Lactose		1.74		
PVP		0.29		
Talc		0.29	2.14	7.22
Eudragit RS PO			1.28	
Eudragit RL100			2.99	
TEC			0.64	
Eudragit E100				22.49
GMS				1.11
Aluminium oxide				1.11
Ethanol		√		√
Isopropanol			√	
Water			√	√

<b>% of coating</b>	<b>3.19</b>	<b>7.05</b>	<b>31.93</b>
---------------------	-------------	-------------	--------------

399 \* The percentages of water in the coating dispersions were 13% and 6% for the second and third coating layers,  
400 respectively.

401

402 It is important to underline that the coating process used for the production of the multi-layered  
403 sustained-release pellets is much simpler and faster than that used in a previous paper for obtaining  
404 omeprazole delayed-release pellets (three coating layers versus five coating layers). This is because  
405 in his case, there is no need to apply two additional isolating layers to protect the active ingredient  
406 from the enteric polymer (which is acidic) or to avoid interaction between the modified-release and  
407 protective coating layers (both are cationic polymers, in contrast to the anionic and cationic  
408 polymers used for omeprazole).

409

### 410 *3.2 Physicochemical characterization of coated pellets*

411 The particle-size distribution of both budesonide formulations was evaluated at the end of each  
412 coating step. Film coating is a complex process because many variables are involved, and the  
413 altering of these parameters is generally restricted [33]. The critical process variables that influence  
414 the coating efficiency include the following: the inlet air temperature and humidity; the air pressure  
415 used for atomization; the flow rate of coating suspension sprayed and of the drying air; and the  
416 possible subsequent curing process [34-36]. Coating experiments were performed to find the right  
417 balance among all these parameters to avoid some common problems met during the coating of  
418 small particles. Such problems can be agglomeration phenomena, spray-drying effects, occlusion of  
419 the nozzle or filter and the non-uniform application of coating dispersion droplets. The flow rate of  
420 the coating suspension and the air pressure defined the spray frequency and the size of droplets  
421 sprayed. Moreover, good ventilation guaranteed a homogeneous flow of particles into the Würster  
422 over a number of applications until the desired coating weight was deposited. Factors such as a  
423 higher coating suspension flow rate, higher humidity or lower drying temperature in comparison to  
424 the optimal values lead to an over-wetting of pellets that can potentially cause agglomeration

425 phenomenon. The generation of electrostatic charges during the coating process is another critical  
426 parameter that can alter the formation of a homogeneous layer on the particles. This is because the  
427 particles remain attached at the wall of the main chamber, and so avoid receiving the sprayed  
428 coating suspension, which dries and flows away through the filter.

429 The generation of electrostatic charges was particularly high with the outermost protective coating,  
430 which contained Eudragit<sup>®</sup> E100. This effect has most influence on this polymer in comparison to  
431 the other polymers. However, as observed from the data shown, the particle size increased during  
432 the coating procedure (Tables 5 and 6, from 277  $\mu\text{m}$  for the neutral pellets to 350-370  $\mu\text{m}$  for the  
433 sustained-release and colon-targeted multi-layered pellets, respectively) thanks to the coating  
434 deposited. However, it remained low enough not to create a swallowing impairment for patients  
435 ( $D(50) < 500 \mu\text{m}$ ) [19]. The low standard deviation values obtained for each coating step give a  
436 good indication of the homogeneity of the coating on the pellets, underlining an effective dispersion  
437 of coating on the particles.

438 At the end of the whole coating process, the mean budesonide content inside the coated pellets of  
439 both formulations was around 90% w/w in comparison to the theoretical loaded drug. This  
440 relatively low drug content is probably due to the loss of a fraction of the sprayed suspension by  
441 spray-drying. The results presented in Tables 5 and 6 showed that the production could be  
442 considered satisfactory for lab-scale coating equipment for such small particles.

443 High levels of residual solvent in the inner structure of coated pellets may compromise their  
444 stability over time during storage. The production process could be considered as satisfactory as the  
445 value of residual solvent reached was limited after the entire coating process and remained below  
446 5% w/w. The coated pellets were analysed by TGA by heating them to temperatures in the range of  
447 between 30°C and 150°C, at which solvents used in the preparation of the coating suspensions were  
448 fully removed, considering their boiling temperatures (78.4°C for ethanol, 100°C for water and  
449 82.5°C for isopropanol) (Table 5 and 6). The obtained results for the residual solvent content (i.e.

450 between 2.4 and 4.5%) demonstrate that these values can be limited once the right parameters are  
 451 set during the whole coating process, reducing the possibility of poor stability of the multi-layered  
 452 pellets during storage in the reconstitutable syrup.

453  
 454 Table 5: Percentage of agglomerates, particle size distribution, yield and residual solvent content results  
 455 obtained after each step of the coating procedure for a representative budesonide colon-targeted multi-  
 456 layered particle batch produced with the selected formulation (D(50) of uncoated pellets<sup>®</sup>:  $277.2 \pm 1.8$ )

457

Colon-targeted formulation	% of agglomerate (n = 1)	Mean diameter D(50) ( $\mu\text{m}$ ) (n = 3, mean $\pm$ s.d)	% of yield (n = 5, mean $\pm$ s.d)	% residual solvent (n = 1)
Budesonide 1 <sup>st</sup> coating	0.0	$278.2 \pm 0.6$	$95.2 \pm 1.0$	4.5
Budesonide 2 <sup>nd</sup> coating	0.2	$315.6 \pm 0.3$	$91.2 \pm 2.6$	2.4
Budesonide 4 <sup>th</sup> coating	0.5	$371.0 \pm 1.5$	$90.1 \pm 4.9$	2.9

458

459 Table 6: Percentage of agglomerates, particle size distribution, yield and residual solvent content results  
 460 obtained after each step of the coating procedure for a representative budesonide sustained-release multi-  
 461 layered particle batch produced with the selected formulation (D(50) of uncoated pellets<sup>®</sup>:  $277.2 \pm 1.8$ )

Sample: sustained-release formulation	% of agglomerates (n=1)	Mean diameter D(50) ( $\mu\text{m}$ ) (n = 3, mean $\pm$ s.d)	% of yield (n = 5, mean $\pm$ s.d)	% residual solvent (n = 1)
Budesonide 1 <sup>st</sup> coating	0.0	$281.1 \pm 0.3$	$92.2 \pm 2.2$	4.5
Budesonide 2 <sup>nd</sup> coating	0.4	$291.2 \pm 0.5$	$91.9 \pm 4.2$	4.4
Budesonide 3 <sup>rd</sup> coating	0.6	$350.5 \pm 6.8$	$91.8 \pm 1.8$	2.8

462

463 *3.3 Microstructure evaluation of coating layers*

464 Generally, uncoated pellets presented a rough and irregular surface [33]. From observation, the  
465 uncoated surfaces (Figure 1) were smooth but irregular and became more rounded, with no visible  
466 cracks, when coating polymers were applied to the surface of the particle using appropriate  
467 experimental parameters. Structural imperfections, such as porosity or discontinuity of the coating  
468 layer and lower layer thickness, could be observed from the cutting of pellets. Such imperfections  
469 should be avoided by optimizing the coating procedures to obtain a modified release of the drug,  
470 with good reproducibility. The surfaces and inner structures of the coated pellets were evaluated by  
471 SEM after the whole coating procedure for both formulations (Figures 1 and 2).

472 As can be seen in Figure 1, the neutral pellets used presented quite smooth surfaces. The  
473 multilayered pellets obtained at the end of each coating step maintained their regular aspect with no  
474 apparent porosity or defects, which indicated that the coating processes were performed  
475 appropriately. Moreover, it should be pointed out that the film formation process in organic solvent-  
476 based systems is fundamentally easier in comparison to that with aqueous-based systems. The  
477 polymer solutions undergo a sol to gel transition upon solvent evaporation to eventually form the  
478 polymeric film [37]. Once the solutions are sprayed onto the pellet surfaces, the organic solvent  
479 evaporates and the polymer chains approach each other to form a thin homogeneous film. As a  
480 consequence of the easier film formation and low initial viscosity presented by the coating  
481 dispersions, a smooth film on particles was provided. Indeed, all the sprayed suspensions presented  
482 relatively low initial viscosities. Consequently, the surface of the film coated pellets was not porous  
483 and the agglomeration phenomena were very limited when spraying the suspension in the fluidized  
484 bed equipment (Figure 1; Tables 5 and 6, agglomeration below 1%).

485 The low percentage of coating sprayed on and the low amount of drug loaded reduced the whole  
486 coating procedure time (< 10 hours). Indeed, the coating time was much shorter to that used for the  
487 preparation of omeprazole pellets (i.e. 5 coating layer, coating time around 24 hours) [22]. This  
488 reduction limited the number of particle-particle and particle-wall collisions, which are normally

489 responsible for small cracks on the particle surface [38]. The major problem related to the coating  
490 process was the generation of electrostatic charges when the polymer employed was Eudragit®  
491 E100. These charges were mainly critical for the external protection coating layer. During this step,  
492 pellets tended to remain attached to the walls of the central chamber. Consequently, they avoided  
493 the appropriate application of the coating droplets onto their surface and their size increased, both  
494 of which facilitated the spray drying effect. The generation of electrostatic charges was partially  
495 solved thanks to the addition of water in the coating dispersions, which increased the humidity  
496 inside the main chamber. Imperfections such as film porosity, resulting from spray drying  
497 phenomenon, and cracks, resulting from multiple particle-particle and particle-wall collisions or  
498 from the formation of repetitive particle-particle sticking detachment [38], can be observed as  
499 shown in Figure 2. For these small particles, the presence of very small imperfections could  
500 potentially make the protective film much more sensitive to the liquid vehicle during storage, which  
501 may potentially alter the release profiles during the dissolution test.

502

503

### 504 3.4 *Dissolution test*

505 Dissolution tests were performed on both budesonide formulations to demonstrate the achievement  
506 of the release targets, using the methods described in Section 2.3.6.

#### 507 3.4.1 *Budesonide colon-targeted formulation*

508 The requirements imposed by European Pharmacopoeia 9th Edition for solid delayed-release  
509 formulations allow a maximum of 10% w/w of API to be released during the two-hour acidic step  
510 and a minimum of 85% w/w of API to be released during the 45-minute buffer step [39]. As can be  
511 seen in Figure 3, the release of budesonide from the colon-targeted formulation after 2 hours in  
512 acidic medium was below the LOQ, demonstrating that the gastro-resistance of the coating was  
513 preserved at pH 1.2. Moreover, around 95% of budesonide was released at pH 7.5 after 45 min.



514 (Fig. 3). This demonstrated that the last protective layer was dissolved in acid medium during the  
515 first two-hour acidic step and that the colonic layer was properly dissolved at pH 7.5. Indeed, the  
516 last protective layer maintained the stability of the multi-layered microparticles once suspended in  
517 the syrup at pH higher than 5. However, it dissolved quickly at acidic pH. This exposed the layer  
518 responsible for the colonic release, which, once the pH was brought to 7.5, dissolved and released  
519 the active substance. As described above, the pH of the buffer medium was chosen to allow the  
520 dissolution of the polymer selected for the colonic targeting, which happens at pH higher than 7.0.  
521 The same colon-targeted formulation with two different percentages of second coating (17.2% and  
522 24.9% total solids, equivalent to 12.4% and to 18.1% of Eudragit S100) was evaluated to verify the  
523 influence of the coating thickness on the release. In both cases, the targeted release was reached  
524 with similar results (% budesonide released: < LOQ in the acidic step and around 95% in the buffer  
525 step). However, the formulation with the lower percentage of sprayed coating was selected due to  
526 its lower processing time, which is more compatible with a future scaling up.

527 Moreover, similar percentages of released budesonide were obtained when the dissolution test was  
528 performed from the multi-layered pellets dispersed in the syrup, immediately after its reconstitution  
529 (Figure 3). The amount of drug released, in both acid and phosphate buffer medium, corresponded  
530 to the results obtained for dry budesonide multi-layered particles (% budesonide released: < LOQ in  
531 the acidic step and around 95% in the buffer step). This demonstrated that the fourth protective  
532 coating layer presented an appropriate barrier against the external aqueous vehicle.

533

#### 534 3.4.2 Budesonide sustained-release formulation

535 For the sustained-release formulation, the aim was to evaluate the ability of such small coated  
536 pellets to sustain the release of the drug and to modulate the dissolution profile. Modification would  
537 be due to different ratios of the two insoluble polymers (Eudragits<sup>®</sup> RS and RL), as the polymers  
538 have different permeability characteristics. As a first step, different batches were produced up to the

539 second coating, which is responsible for the sustained release. This was done to determine the  
540 appropriate RS-RL ratio to control the budesonide release up to around 24 hours, by which time  
541 most of the API would be released (Fig. 4).

542 Maximal percentages (100%) of both RS PO and RL 100 were used to evaluate the two extreme  
543 dissolution profiles to obtain the lowest and the highest amounts of drug released. To determine the  
544 suitable polymer ratio from the ones produced, dissolution tests were done at a constant buffered  
545 pH of 7.5. This avoided the starting acidic step, as the permeability of both insoluble polymers is  
546 pH-independent and there was no protective coating layer. The dissolution tests were performed  
547 over 24 hours even though the residence in the colon after the transit in the stomach and in the small  
548 intestine is within a variable period of time (between 7 to 48 hours) [40]. Indeed, it seemed  
549 reasonable to consider a maximum dissolution test duration of 24 hours as the GI transit times are  
550 generally shorter for small particles, especially when using a liquid dosage form.

551 The examination of the dissolution results obtained in Figure 4 showed that the drug release was  
552 very low when the pellets were coated with Eudragit RS alone (around 5% release within 24h,  
553 showing that the film permeability was too low). On the other hand, the budesonide release was  
554 around 80% within 24h for the pellets coated with Eudragit RL alone. As expected, when the ratio  
555 of the high-permeability polymer RL was increased in the polymer blends, the drug release profiles  
556 increased from around 40% (RS:RL 70:30) to around 78% (RS:RL 30:70) of drug released within a  
557 24h dissolution test. When the polymers were present in an equal blend ratio (RS:RL 50:50), the  
558 drug release was around 55% within 24h.

559 The RS:RL 30:70 ratio was selected because its ability to release the API was similar to that  
560 obtained with the high permeability polymer alone (RL 100%) and consequently higher in  
561 comparison the other ratios. Moreover, the presence in this ratio of 30% of the low-permeability  
562 polymer (RS) guaranteed a better protection of multicoated pellets once dispersed in the syrup,  
563 during storage.

564 Two different percentages of the second coating layer (7.1% and 9.5% total solids, equivalent to  
565 4.3% and to 5.8% of polymer blend), using a RS:RL ratio of 30:70, were then evaluated to verify  
566 the influence of the coating thickness on the drug release. It was demonstrated that the increase in  
567 the thickness of the second coating layer from 7.1% to 9.5% provoked a decrease in the release of  
568 budesonide within the 24h dissolution test from 78% to 68%, respectively (data not shown).  
569 Consequently, two batches with the two different percentages of the second coating (i.e. 7.1% and  
570 9.5% of total solids) were produced until the last protective coating layer. These were analysed in  
571 terms of dissolution profile and stability after dispersion in the syrup. Once the final batches were  
572 produced, dissolution tests involving two hours of acidic step at pH 1.2 and 22 hours of buffer step  
573 at pH 7.5 (with withdrawals after 1, 2, 4, 6, 8 and 22 hours from the beginning of the buffer step)  
574 were performed.

575 In both cases, as the progressive release of the drug was effective from the beginning of the  
576 dissolution test, the outermost layer was quickly dissolved without an initial lag time within the first  
577 two hours of the acidic step. As soon as the last coating was dissolved, the sustained-release layer  
578 started to control the release of budesonide (Fig. 5). Moreover, the release of the drug continued  
579 after the pH changed (from 1.2 to 7.5), following the two-hour acidic step. The release of  
580 budesonide from the coating consisting of two insoluble polymers, was pH-independent.

581 The comparison of the release profiles obtained from the multi-layered pellets before (dry) and after  
582 dispersion in the syrup permitted the observation that slightly higher percentages of budesonide  
583 were released when the dissolution test was performed on the multi-layered pellets dispersed in the  
584 syrup, immediately after its reconstitution (Fig. 5). The amount of drug released from the  
585 formulation with the two different coating percentages (2<sup>nd</sup> coating equal to 7.1% and 9.5%)  
586 occurred in similar quantities, reaching values higher than 90% w/w in 24 hours. This result  
587 demonstrated that the fourth coating was an appropriate barrier against the external aqueous vehicle

588 in both cases. As such a thick coating layer is more compatible with a future scaling up in industry,  
589 the formulation with a lower percentage of coating (2<sup>nd</sup> coating equal to 7.2%) was selected.

590 The similarity factor  $f_2$  was calculated for this formulation for the different time points (1 and 2  
591 hours in the acidic step and 1, 2, 4, 6, 8 and 22 hours in the buffer step). This value was in the range  
592 between 50 and 100. This showed that no significant modification in terms of the sustained-release  
593 properties of the second coating were observed after dispersion of multicoated pellets in a neutral  
594 pH aqueous medium.

595

596

### 597 *3.5 Storage stability test*

598 The reconstitutable syrup exhibited its capability to provide a stable system for at least 2 weeks at  
599 ambient temperature for both formulations of budesonide multi-layered particles. As shown in  
600 Table 7, after a storage period of 2 weeks in the syrup, a minimum of 89% w/w of the budesonide  
601 nominal dose was released from the colon-targeted coated pellets in the 45 minutes of buffer stage.  
602 Indeed, it was observed that the release of BUD in the buffer medium decreased slightly during the  
603 storage. Release went from 97.8% immediately after dispersion to 89.3% after 2 weeks of storage at  
604 25°C. To verify the reason for the decrease in BUD release after storage, the dissolution test was  
605 prolonged to 4 hours in the buffer medium. From this analysis, it was clear that the drug release was  
606 not completed within 45 min in the buffer stage, or when the storage time was increased. This could  
607 be due to the progressive swelling and diffusion of the polymers used in the different coating layers  
608 when the pellets were dispersed in the reconstituted syrup. Indeed, an interaction between the  
609 cationic and the anionic polymer (E100 and S100) can occur during storage, forming a less  
610 permeable coating that can alter the diffusion of the dissolved drug. The results in Table 7 showed  
611 that the decreasing effect observed during storage in the reconstituted syrup was higher after 2  
612 weeks of storage and greater release of the drug after 4h dissolution in the buffer stage.

613 The  $p$  was calculated between the average of the dissolution points after 45 minutes of buffer step,  
 614 (the point to be considered in accordance with the European Pharmacopoeia 9<sup>th</sup> Edition) at time  
 615 zero and at the stability at 1 week, using the Student t-test. The results demonstrated that the two  
 616 sets of data were not significantly different from each other ( $p=0.07$ ). When the  $p$  was calculated  
 617 between the average of dissolution points, after 45 minutes in the buffer step, at time zero and at the  
 618 stability at 2 weeks, the results showed a significant difference between the two sets of data  
 619 ( $p=0.003$ ). Nevertheless, the drug release values obtained for up to 2 weeks' storage satisfied the  
 620 common requirements for delayed-release dosage forms. Less than 10% of BUD was released in the  
 621 2 hours of the acidic stage and more than 85% of BUD was released in the first 45 minutes of the  
 622 buffer stage, considering all the storage periods up to 2 weeks.

623 Table 7: Percentages of budesonide released from colon-targeted multicoated pellets dispersed in the reconstituted  
 624 syrup at the predetermined time points of storage at 25°C (time zero, 1 and 2 weeks): 2-hour acidic stage (pH 1.2),  
 625 followed by a pH 7.5 buffer stage at different time points: 45 minutes, 2 and 4 hours (n=6, mean  $\pm$  s.d.)

Time points:	% BUD released after 2 h in the acidic stage	% BUD released after 45 min. in the buffer stage	% BUD released after 2h in the buffer stage	% BUD released after 4h in the buffer stage
Time zero	0.96 $\pm$ 0.11	97.8 $\pm$ 1.8	101.3 $\pm$ 1.2	100.5 $\pm$ 1.5
1 week of storage	2.81 $\pm$ 0.24	95.5 $\pm$ 4.2	101.8 $\pm$ 5.0	101.6 $\pm$ 3.6
2 weeks of storage	4.25 $\pm$ 0.37	89.3 $\pm$ 4.2	90.8 $\pm$ 3.7	96.5 $\pm$ 3.6

626

627 In the case of the sustained-release formulation, the reconstituted syrup also exhibited its capability  
 628 to provide a stable system for budesonide multi-layered particles with the selected formulation, with  
 629 7.1% second sprayed coating layer. Indeed, after a storage period of 2 weeks in the syrup, the drug  
 630 release obtained at the end of the dissolution test in the buffer medium was limited and decreased  
 631 from 90.2% to 87.2% w/w of budesonide's nominal dose (Table 8). A deeper evaluation of the  
 632 dissolution results obtained during storage showed that the decrease in drug released is higher  
 633 during the first hours of the dissolution test (e.g. budesonide release values of 60.0%, 52.9% and  
 634 49.7% after 2 h in the buffer medium for the coated pellets immediately after reconstitution, and  
 635 after 1 and 2 weeks of storage, respectively). These slight modifications in the release profiles could

636 be explained by the potential diffusion of the aqueous medium through the different coating layers  
 637 and their progressive swelling during the storage in the syrup. Both the compositions of the  
 638 protective layer (i.e. addition of hydrophobic compounds) and the syrup (i.e. addition of high  
 639 concentrations of sorbitol and of viscosifying agents) were chosen to try to limit the penetration of  
 640 water into the pellets and to guarantee the stability of the pellets after reconstitution.

641 The similarity factor ( $f_2$ ) was calculated for all the dissolution results generated to verify the  
 642 similarity of the dissolution profiles. The  $f_2$  values obtained were 65% and 60% (in comparison to  
 643 the initial dissolution profiles), after 1 week and 2 weeks, respectively. Therefore, the dissolution  
 644 profiles obtained during the 2-week storage at 25°C could be considered similar to that at time zero.

645 The stability of the dissolution was also tested for the multilayered batch with the higher percentage  
 646 of second coating (9.5%). In this case, a good stability of the coated pellets was provided after their  
 647 dispersion in the syrup for the two weeks of storage, although with a lower percentage of  
 648 budesonide released in a 24 h dissolution test (maximum release in the 24h of buffer step equal to  
 649  $91.0 \pm 1.4$  at time 0,  $89.1 \pm 1.7$  at 1 week and  $81.4 \pm 1.1$  at 2 weeks).

650 Table 8: Percentages of budesonide release from sustained-release multicoated pellets with 7.1% of second coating total  
 651 solids (equivalent to 4.3% of polymers blend), dispersed in the reconstituted syrup at predetermined time points of  
 652 storage at 25°C (time zero, 1 and 2 weeks): 2-hour acidic stage, followed by a 24-hour pH 7.5 buffer stage (n=6, mean  
 653  $\pm$  s.d.)

Samples	Time zero	% drug release	
		1 week of storage	2 weeks of storage
1 h acid pH 1.2	$35.0 \pm 1.8$	$27.6 \pm 0.7$	$24.9 \pm 0.7$
2 h acid pH 1.2	$48.4 \pm 2.7$	$39.6 \pm 1.7$	$36.8 \pm 1.9$
1 h buffer pH 7.5	$51.7 \pm 1.9$	$40.3 \pm 2.5$	$37.2 \pm 1.9$
2 h buffer pH 7.5	$60.0 \pm 1.9$	$52.9 \pm 0.9$	$49.7 \pm 1.4$
4 h buffer pH 7.5	$70.9 \pm 2.6$	$62.0 \pm 3.3$	$56.6 \pm 0.9$
6 h buffer pH 7.5	$78.3 \pm 3.2$	$64.0 \pm 1.6$	$61.3 \pm 1.8$
8 h buffer pH 7.5	$81.9 \pm 3.1$	$72.7 \pm 1.5$	$69.5 \pm 1.5$
24 h buffer pH 7.5	$90.2 \pm 4.2$	$89.4 \pm 2.9$	$87.2 \pm 2.1$

654

655

**656 4. Conclusion**

657 In this study, the oral system based on a multi-layered particle technology showed an innovative  
658 process that will permit the production of reconstitutable liquid dosage forms with modified-release  
659 properties. The multi-layered microparticles obtained using a coating procedure in a bottom-spray  
660 fluid-bed coater maintained, at the end of the process, the smaller size which has been highly  
661 awaited to address the need in patients with swallowing impairment, such as paediatric, elderly or  
662 critically ill patients. Such small particles dispersed in a syrup could improve patients' compliance  
663 and consequently the effectiveness of their therapy.

664 In this study, it was demonstrated that the multi-layered particle technology is a flexible  
665 manufacturing process that is suitable for different kinds of release, such as colon-targeted or  
666 sustained release. In both the developed formulations, the multicoated pellets were able to provide  
667 an appropriate control of the drug release in both dry and liquid dispersed states. Moreover, this  
668 new approach showed a good stability of the drug release profiles for at least 2 weeks when the  
669 multi-layered pellets were suspended in a syrup and stored at ambient temperature. From an  
670 industrial perspective, it could be interesting to apply this kind of technology to other drugs.

671

**672 Role of the funding source**

673 This work was financially supported by BePharBel Manufacturing. The sponsor of this study  
674 approved the study design, but was not involved in the interpretation of the data or in the writing of  
675 the article. The sponsor has read and approved the content of this article before its submission.

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679 **References**

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1 Fig. 1: SEM pictures of budesonide colon-targeted and sustained-release batches surface at the end  
2 of a coating procedure, taken following the formulation and coating parameters presented in Tables  
3 1-4, respectively. The microstructure of the coated pellets was compared with the neutral pellets  
4 (Cellets<sup>®</sup> 263), and analysed at different magnification (250x, 500x, 1000x).

5

6

7 Fig. 2: SEM pictures of the cross section of budesonide colon-targeted and sustained-release  
8 batches at the end of a coating procedure, taken following the formulation and coating parameters  
9 presented in Tables 1-4, respectively. The coated pellets were analysed at different magnification  
10 (500x, 1000x). Imperfections such as porosity and small cracks are shown.

11

12

13 Fig. 3: Dissolution profiles of the colon-targeted multi-layered pellets (17.2% of 2<sup>nd</sup>-coating  
14 total solids, equivalent to 12.4% of Eudragit S100) before and after dispersion in a conventional  
15 syrup, after 2 hours in the acidic stage and after 45 minutes in the pH 7.5 buffer stage (mean  
16 values +/- s.d., n=6).

17

18

19 Fig. 4: Influence of the different ratios of insoluble polymers (RS:RL) in the modified-release  
20 layer on the dissolution profiles of budesonide, done in a pH 7.5 buffered medium. The test was  
21 performed on the dry sustained-release pellets coated up to the second layer, which consisted of  
22 9.5% of total solid, equivalent to 5.85% of insoluble polymer blends (mean values +/- s.d.,  
23 n=6).

24

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26 Fig. 5: Dissolution profiles of the dry sustained-release multi-layered pellets before and after  
27 dispersion in a conventional syrup after 2 hours in the acidic stage pH 1.2 and 22 hours in a pH  
28 7.5 buffer medium (mean values, n=6). The coated pellets tested were constituted of the same  
29 formulation as shown in Table 4 but with two different percentages of 2<sup>nd</sup> coating layer: 7.1%  
30 and 9.5% of total solids, equivalent to 3.4% and to 5.8% of polymer blends using an RS/RL  
31 ratio of 30:70.

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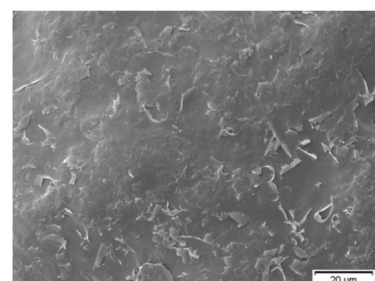
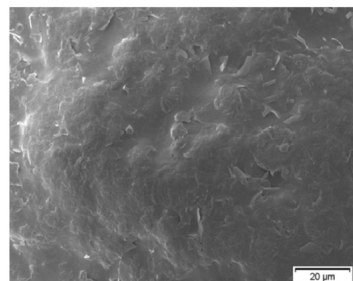
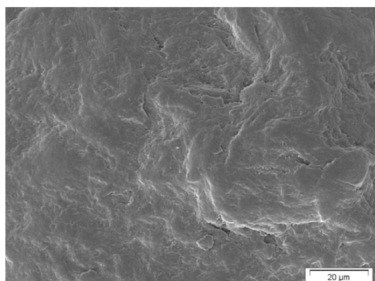
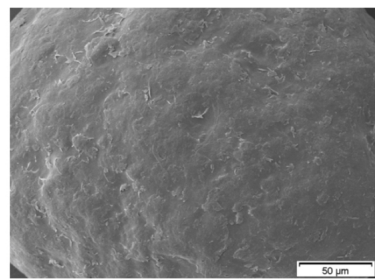
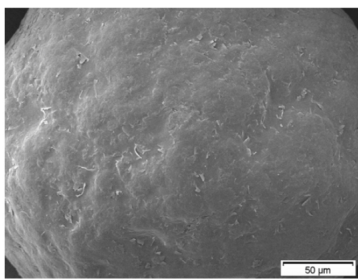
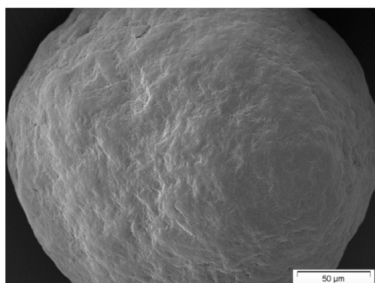
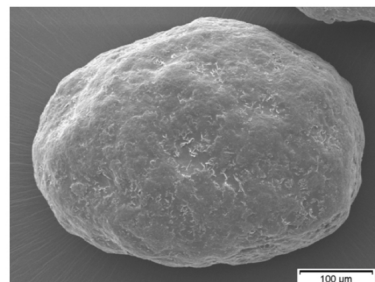
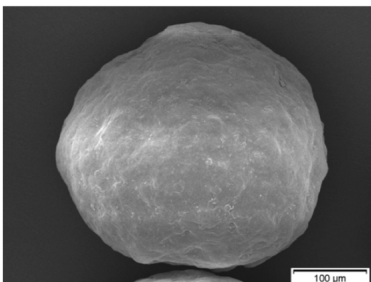
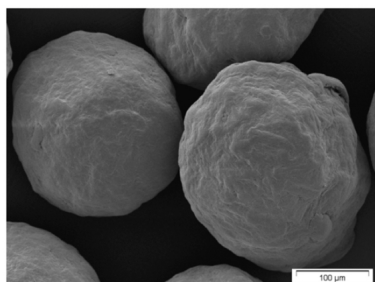
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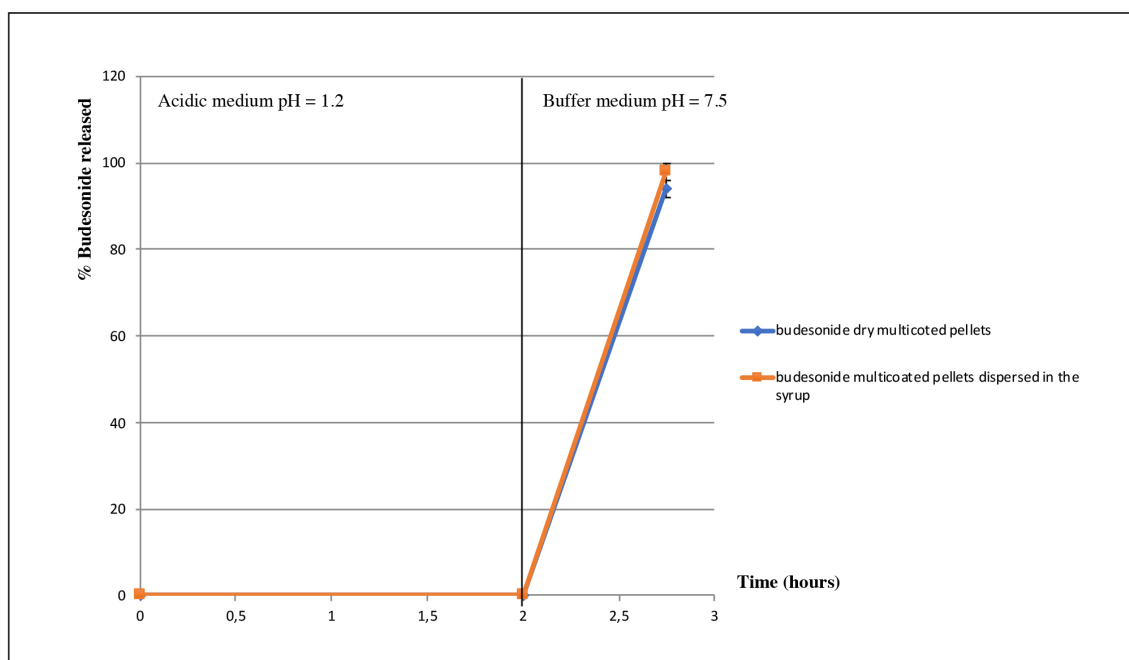
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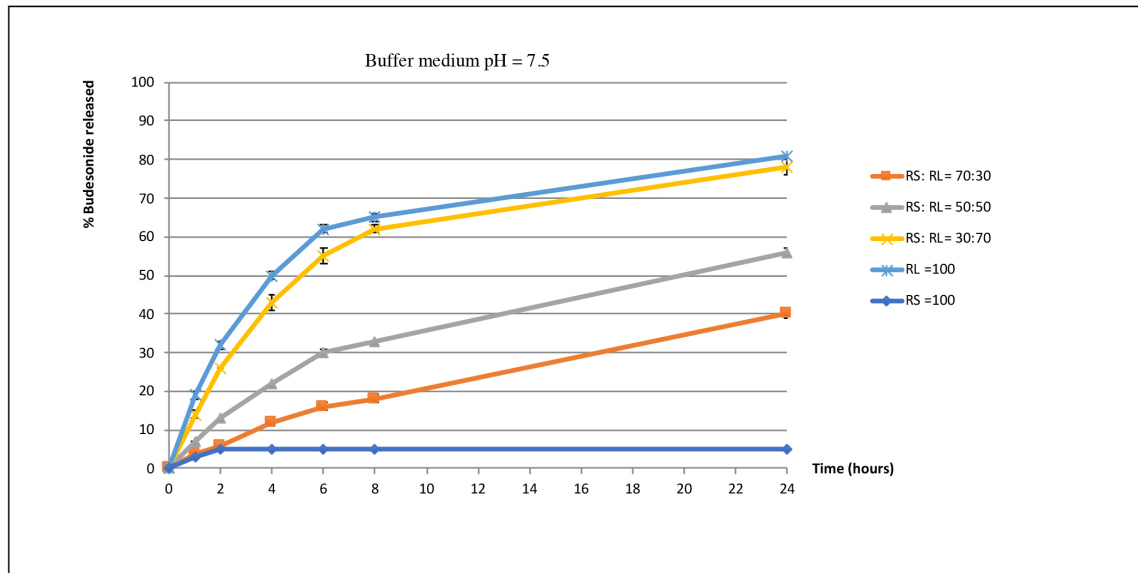
Neutral pellets (Cellets® 263)

Budesonide colon-targeted formulation

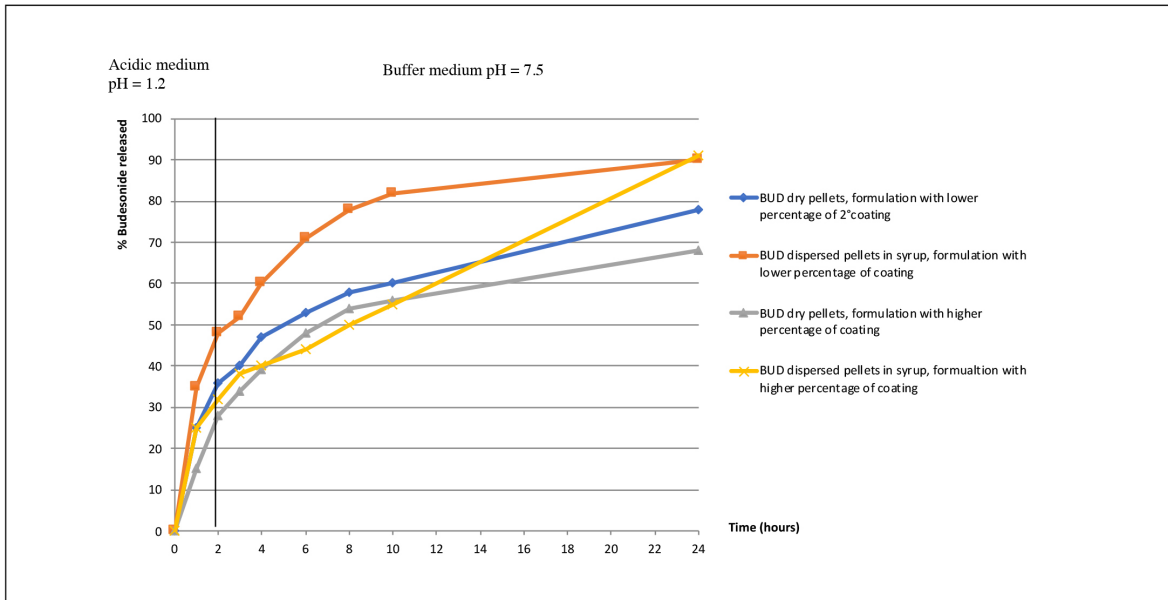
Budesonide sustained-release formulation





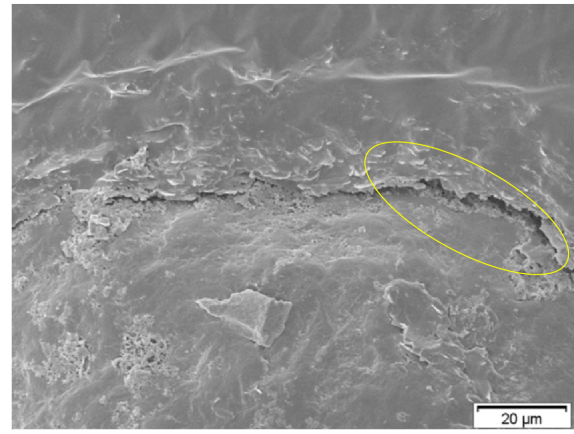
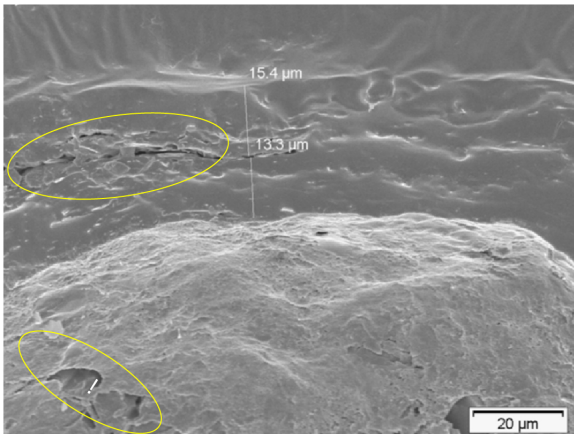
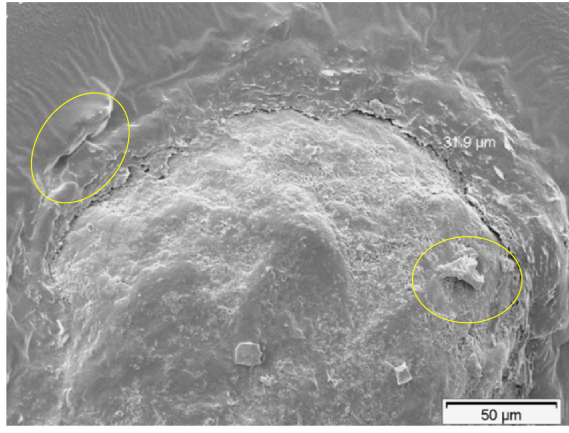
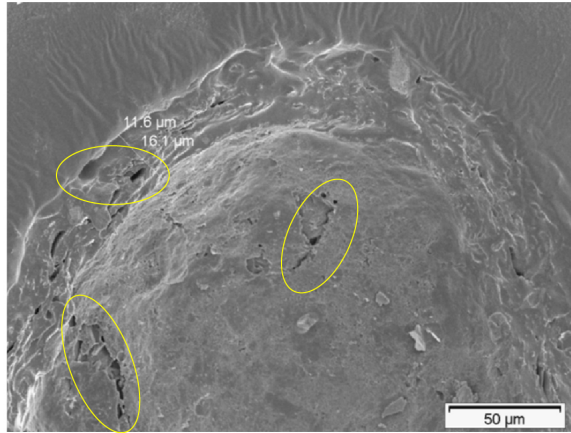






Budesonide colon-targeted formulation

Budesonide sustained-release formulation



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**Declaration of interests**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

This work was financially supported by BePharBel Manufacturing. The sponsor of this study approved the study design but it was not involved in the interpretation of the data or in the writing of the article. The sponsor has read and approved the content of this article before its submission.