1	Effect of the Nitrogen Source on the Hydrogen Production Metabolism and
2	hydrogenases of Clostridium butyricum CWBI1009
3	Christopher Hamilton ¹ , Magdalena Calusinska ³ , Stacy Baptiste ¹ , Julien Masset ¹ ,
4	Laurent Beckers ¹ , Philippe Thonart ¹ , Serge Hiligsmann ^{1,2*}
5	¹ Walloon Centre for Industrial Biology, Boulevard du Rectorat 29, B4000
6	Liege, Belgium
7	² 3BIO-BioTech, Brussels School of Engineering, Université Libre de Bruxelles,
8	Avenue F. Roosevelt, 50, CP165/61, B1050 Brussels, Belgium
9	³ Centre for Protein Engineering, Bacterial Physiology and Genetics, University
10	of Liege, Allée de la Chimie 3, B4000 Liege, Belgium
11	*Corresponding author : serge.hiligsmann@ulb.ac.be ; +3226503049
12	Abstract
13	Investigations were carried out to determine the effect of various concentrations
14	of organic and ammonium nitrogen sources on fermentative hydrogen
15	production by the strain Clostridium butyricum CWBI1009. The results
16	indicated that the H ₂ -producing metabolism of the strain is favoured within the
17	range (0.56-0.062 g_N/L) of peptone and (NH ₄) ₂ SO ₄ . Optimal overall
18	
	performance (i.e. 1.43 ± 0.08 mol H ₂ /mol glucose and 1.08 ± 0.03 mL _{H2} /h,
19	performance (<i>i.e.</i> 1.43 \pm 0.08 mol H ₂ /mol glucose and 1.08 \pm 0.03 mL _{H2} /h, respectively) was achieved with 0.062 g _N /L of casein peptone. The study of the
19 20	performance (<i>i.e.</i> 1.43 \pm 0.08 mol H ₂ /mol glucose and 1.08 \pm 0.03 mL _{H2} /h, respectively) was achieved with 0.062 g _N /L of casein peptone. The study of the amino acid uptake and the gene expression pattern for four [FeFe]-hydrogenases
19 20 21	performance (<i>i.e.</i> 1.43 \pm 0.08 mol H ₂ /mol glucose and 1.08 \pm 0.03 mL _{H2} /h, respectively) was achieved with 0.062 g _N /L of casein peptone. The study of the amino acid uptake and the gene expression pattern for four [FeFe]-hydrogenases and the nitrogenase showed that nitrogen was in excess in all the experiments

with a nitrogen concentration above $0.062 \text{ g}_{\text{N}}/\text{L}$ and, at that optimal concentration, the expression of the *HydB2* gene would be responsible for the much higher H₂ yield.

Keywords : biohydrogen; *Clostridium*; nitrogen; hydrogenase; amino acid
 26

27 1. Introduction

Hydrogen is regarded as an attractive energy vector for the future due to its 28 clean and direct combustion characteristics. Moreover, a hydrogen-based 29 30 economy could reduce our dependence on fossil fuels. However, hydrogen is mainly found as part of molecular compounds and currently its production is 31 32 largely dependent on the use of fossil fuels such as natural gas. This results in the release of huge amounts of non renewable CO₂ and consequently to global 33 warming and environmental degradation [1, 2]. Alternatively, hydrogen 34 35 produced by microorganisms from renewable biomass is promising since the whole process can be considered as carbon neutral [3]. 36

37

Among the different microbiological routes to hydrogen, dark fermentation using facultative and/or strict anaerobic bacteria has received major attention. Most of the extensive research on this topic has focussed on the advantages of hydrogen production combined with waste treatment [4-7] since the major

limiting factor in the scaling up of dark fermentation is the high cost of the main 42 feedstock and some other nutrients to promote growth and metabolism. 43 Pure sugars are frequently used as the carbon source and casein peptone or yeast 44 45 extract as the organic nitrogen source. For industrial-scale applications they should be replaced by readily available and cheap carbohydrate-rich residues 46 and ammonia-based substitutes in order to decrease process costs [2, 6, 8]. 47 48 Nitrogen-containing nutrients (e.g. ammonia, vitamins, and proteins) have a critical impact on fermentation performance since nitrogen is required for cell 49 50 replication, maintenance, metabolism, and the production of carbohydrate-51 hydrolysing enzymes [9, 10]. Moreover, the literature clearly indicates that 52 acidogens such as *Clostridium sp.* and *Enterobacter sp.* can utilise not only organic nitrogen (e.g. proteins, amino acids), but also ammonia [9, 11, 12]. 53 Therefore, the use of suitable pretreated residues containing ammonia (e.g. 54 animal manure) should also be considered in biohydrogen production, since it is 55 56 widely available and its disposal represents a significant cost if serious risks of air and water pollution are to be avoided [13]. As a consequence, there is a need 57 to deeply investigate the nitrogen requirements for H₂-producing strains and 58 59 their impact on bacterial metabolism and major enzymes. The Clostridium butyricum CWBI1009 strain was selected for the study due to 60

60 The *Clostridium butyricum* CWBI1009 strain was selected for the study due to
61 its high hydrogen production activity while fermenting different carbon sources
62 (glucose, disaccharides or starch) [14-16]. Moreover, the strain achieved very

high performance (up to 3.4 mol H₂/mol glucose) in immobilised cultures [17].
Hydrogen is produced via pyruvate-ferredoxin oxidoreductase (PFOR) and, at a
lower extent, via NADH-ferredoxin oxidoreductase (NFOR) [18, 19]. Both
enzymatic complexes transfer electrons to [FeFe]-hydrogenases which generate
hydrogen by a reversible reduction of protons accumulated during fermentation.
This mechanism controls the electron flow within the cell and therefore
determines the direction of by-product synthesis [20].

However, little is known about the physiological and biochemical properties of
clostridial [FeFe]-hydrogenases and their role in the hydrogen metabolism [19,
20] since most of the investigations focussed on the commonly known CpI
(HydA) from *Clostridium pasteurianum* [21]. *Clostridium butyricum*CWBI1009 possesses four different [FeFe]-hydrogenases, three of which
(*hydA2, hydB2, hydB3*) are monomeric intracellular enzymes and one (*hydA8*) is
associated with the heterotrimeric hydrogenase cluster [19, 20].

The objectives of this study were to further optimize the nitrogen source for biohydrogen production by the strain *Clostridium butyricum* CWBI1009. This was carried out by monitoring the H_2 yields (mole of H_2 produced per mole of glucose consumed), production rates (mole of H_2 produced per hour) and the formation of soluble by-products in 270 mL batch cultures at different ammonium and peptone concentrations. In addition, particularly relevant conditions were selected for amino acid fingerprinting (HPLC-UV) and

84	expression pattern analysis (RT-qPCR) of the four [FeFe]-hydrogenase genes
85	using specifically designed primers [22, 23]. The same was done for the
86	nitrogenase gene since its role in hydrogen production by Clostridium butyricum
87	CWBI1009 appears to depend on the environmental conditions [19].

89 2. Materials and methods

90 2.1. Bacterial strain and culture media

The strain cultured in this work, *Clostridium butyricum* CWBI1009 (accession 91 92 no. GU395290), was previously isolated from an anaerobic sludge [14]. The strain was grown in variously formulated buffered synthetic media previously 93 growth and hydrogen production with Clostridium 94 used for and Enterobacteriaceae [11, 14, 16]. The MDT culture medium (without the added 95 variable mineral or organic nitrogen source) contained, per litre of deionised 96 97 water, glucose monohydrate (5g), yeast extract (0.5g), Na₂HPO₄ (5.1g), KH₂PO₄ (1.2g), MgSO₄.7H₂O (0.5g), cysteine hydrochloride monohydrate (0.5g). L-98 99 cysteine hydrochloride was used to ensure strictly anaerobic conditions in the 100 media and its effect was confirmed by measured redox potential ranging from -200 to -250 mV. To determine the impact on hydrogen production of an organic 101 nitrogen source (casein peptone) compared to an ammonia-based substitute 102 $((NH_4)_2SO_4)$, each of the two sources was used in a separate experiment at 10 103 concentrations ranging from 0.025 to 2.3 g of total nitrogen/L added to MDT 104

medium (g_N/L) . They supplement the basal nitrogen content of 0.055 and 0.034 105 g_N/L related to the yeast extract and cysteine initial concentration in MDT 106 medium, respectively. The total amount of nitrogen in the medium compounds 107 108 was calculated according to the supplier's product description and confirmed by measuring the total nitrogen and ammonia content using HACH kits for total 109 nitrogen and ammonia and a HACH DR/2010 spectrometer. All the chemicals 110 111 used were of analytical or extra pure quality and were supplied by Merck, VWR and Sigma. Casein peptone and yeast extract were supplied by Organotechnie 112 113 (La Courneuve, France).

114 2.2. Experimental set-up

BHP (Biochemical Hydrogen Potential) tests were carried out to assess the influence of both nitrogen sources in 270 mL serum bottles (200 mL working volume). The sterile carbon source (glucose monohydrate in solution in deionised water) was added separately (to a final concentration of 5 g/L) to prevent Maillard reactions between the carbohydrates and the amino acids. The L-cysteine hydrochloride monohydrate was also added separately to obtain a final concentration of 0.5 g/L.

After the preparation of the MDT medium and adjustment of the pH to 7.3 the bottles were sterilised and cooled down before inoculation with 3 mL suspension from a fresh (incubated for 48 h) preculture tube. The bottles were then capped with a butyl stopper as described by Lin *et al.* [24] and flushed with nitrogen to remove the remaining oxygen in the gas phase before incubation at
30 °C. Each BHP test was conducted in triplicate at least and all the
experimental conditions were monitored until biogas production had completely
ceased (*i.e.* 96 h with casein peptone and 144 h with ammonium sulfate).

130 **2.3. Monitoring and analytical methods**

131 Biogas analysis

132 The biogas produced in the bottle experiments was collected by means of sterile syringes as described by Hamilton et al. [11]. Water supplemented with 9 N 133 134 KOH was used in replacement equipment designed by Hiligsmann et al. [16] in 135 order to sterilely monitor biogas composition (H₂ and CO₂), except for the last non-sterile gas sample which was analysed by gas chromatography. The 136 137 proportion of hydrogen gas was determined using a gas chromatograph (GC: Hewlett Packard 5890 Series II, UK). The chromatograph was fitted with a 138 thermal conductivity detector (TCD) and a 30 m x 0.32 mm GAS PRO GSC 139 140 capillary column (Altech) in series with a 20 m x 0.25 mm CarboPLOT P7 column (Chrompak). The temperatures of the injection, the TCD chambers and 141 the oven were maintained at 90°, 110° and 55 °C respectively. Nitrogen was 142 143 used as the carrier gas in the column at a flow rate of 20 mL/min.

144 *Glucose and soluble metabolites analysis*

145 Culture samples were collected regularly to operate measurements and to146 harvest culture components. The samples were centrifuged at 12000 g for 30

seconds and the supernatants were filtered through a 0.2 µm cellulose acetate 147 membrane (Sartorius Minisart). The analyses were performed using an Agilent 148 1100 series HPLC and an Agilent ChemStation was used for instrument control 149 150 and data acquisition (Agilent Technologies, CA, USA). For the quantification of glucose, ethanol, lactate, acetate, formate and butyrate, the HPLC was equipped 151 with a Supelcogel C610-H column preceded by a Supelguard H precolumn 152 (oven temperature 40 °C) and a differential refraction index detector (RID, 153 detection cell maintained at 35 °C). An isocratic mobile phase with 0.1% H₃PO₄ 154 (in MilliQ water) was applied at a flow rate of 0.5 mL/min for a duration of 35 155 156 min at a maximum pressure of 60 bars.

157 Amino acids analysis

For the separation of amino acids by reverse phase chromatography, the HPLC 158 was fitted with a Zorbax Eclipse XDB-C₁₈ column (150×4.6 mm, 5 µm, 159 Agilent) preceded by a Phenomenex C_{18} security guard column (4.0 \times 3.0 mm, 160 Phenomenex). The mobile HPLC grade phases A (40 mmol/l Na₂HPO₄ at pH 161 7.8) and B (45% acetonitrile, 45% methanol, 10% water) were filtered through a 162 0.22-micron Millipore Durapore PVDF membrane filter. Automatic pre-column 163 164 derivatisation with 9-Fluorenylmethyl chloroformate (OPA-3MPA) and ophthalaldehyde 3-mercaptopropionic acid (FMOC) was performed at room 165 166 temperature, in accordance with the injector programs listed in Table 1, using 1 μ L of sample. After the derivatisation 0.5 μ L of the mixture was injected for 167

each chromatographic separation. Primary amino acids were derivatised online 168 with OPA-3MPA and identified with a photodiode-array UV detector at 338 nm. 169 Hydroxyproline and proline were derivatised by adding FMOC to OPA-3MPA 170 171 and detected at 262 nm. The chromatographic separation was obtained using the elution gradient indicated in Table 2 with a column temperature of 40 °C. The 172 column was initially conditioned as follows: mobile phase B was pumped for 25 173 174 min at a flow rate of 2 mL/min, followed by 15 min of 10% methanol at 1.0 mL/min and 20 min of mobile phase A (pH 7.8) at 2 mL/min. The flow rate was 175 176 constant at 2.0 mL/min throughout the analysis. The run time, injection to 177 injection, was of 27 min. In order to maintain column integrity, a mixture of 90% isopropanol and 10% methanol was run through it after every 15 injections. 178 179 The linearity and accuracy of the method were assessed using three different concentrations (1, 5 and 10 mg/mL) of amino acid standard mixtures injected in 180 triplicate. All common 20 amino acid standards were obtained from Sigma 181 Chemical. FMOC, OPA-3MPA and borate buffer were obtained from Agilent. 182

183 **2.4. RT-qPCR**

RNA was extracted using a RiboPureTM-Bacteria (Ambion) and cDNA was
prepared as described in the Reverse Transcription System manual (Promega);
as previously described [19]. Specific primers and probes for the 4 [FeFe]hydrogenases of *Clostridium butyricum* CWBI1009, *i.e. hydA2*, *hydA8*, *hydB2*and *hydB3* and *16SrRNA* were designed in this study. Primers targeting the

reference genes, *i.e. recA*, *gyrA* were as described before [23] and are given in
Table 3. The quantitative PCR amplifications were carried out with a Mini
Opticon (BioRad) apparatus.

The DNA template used for a standard curve was prepared as previously 192 described [23]. For the gene expression analysis 1 µL of cDNA was used. The 193 total volume of the PCR mix was 25 µL. Each reaction consisted of 1 x PCR 194 mix (ABsoluteTM Blue QPCR SYBR[®] Green Fluorescein Mix or 195 ABsoluteTMBlue QPCR mix, Thermo Scientific) with each primer or hydrolysis 196 197 probe (HPLC cleaned, Biomers, Germany) at a final concentration of 150 nM. 198 Each sample was analysed in triplicate. A 'no template' control was included in each run. The specificities of the primers were verified at the end of each qPCR 199 reaction by performing a melting curve analysis (SybrGreen based 200 quantification). The standard curve preparation and the cycling conditions for 201 SybrGreen chemistry were as previously described [23]. For the probe-based 202 chemistry the initial denaturation of 15 min was followed by 40 cycles of 203 denaturation at 95 °C for 15 s and a primer annealing/amplification step at 60 °C 204 for 30 s. The reaction efficiency was calculated as factor specific using the 205 equation $E=10^{-1/\text{slope}}$ [25]. 206

For gene expression analysis the relative expression levels were calculated with a Relative Expression Software Tool 2009, REST© [26]. To estimate the upand down-regulation of the genes analysed, the Cqs obtained were compared

with those of the reference genes and with an external standard control gene.
The stability of the 3 reference genes was determined using the
BestKeeperExcel-based tool [27].

- 213
- 214 3. Results and discussion

Very few information is currently available about the impact of different types 215 and concentrations of nitrogen sources on hydrogen production by pure 216 Clostridium butyricum and on the associated by-products [9, 28]. Therefore, the 217 218 experiments reported here were carried out to study the effect of 10 different concentrations of casein peptone (11.2% total nitrogen, average molecular 219 weight 681 Da) and 10 equivalent concentrations of (NH₄)₂SO₄. In order to 220 221 enable easy comparisons, the nitrogen content in the different media was 222 reported in terms of total nitrogen.

223 Aiming to rapidly survey a large range of potentially interesting conditions, 224 BHP tests were carried out in 270 mL serum bottles without pH regulation. No 225 more biogas production could be measured after 96 h of incubation with casein 226 peptone and 144 h with (NH₄)₂SO₄. The hydrogen production evolved according 227 to a sigmoidal curve as reported in former experiments with the same *Clostridium* strain in BHP tests with 5 g/L casein peptone [14, 16]. The H_2 228 yields (mole of H₂ produced per mole of glucose consumed), H₂ production rates 229 (mole of H₂ produced per hour) and the release of soluble by-products were 230

monitored for the 20 different conditions. In order to better understand the large differences that were observed in H₂ production activity, three conditions (0.062 and 0.056 g_N/L casein peptone and 0.56 g_N/L (NH₄)₂SO₄) were selected for a study of (i) the amino acid distribution (HPLC-UV) and of (ii) the gene expression pattern (RT-qPCR) for the four [FeFe]-hydrogenases and for the nitrogenase, all of which had previously been identified in the genome of *Clostridium butyricum* [20] using specifically designed primers [19, 23].

3.1. H₂ production activity

250

239 *Casein peptone as nitrogen source*

240 The control sample (indicated by a black arrow on Fig. 1) contained the casein peptone concentration used in previous laboratory studies, *i.e.* 5 g/L containing 241 242 0.56 g_N/L total nitrogen [11, 14]. With this sample, a total of 181.5 ± 7.79 mL_{biogas} were produced after 96 h of culture and, due to release of acid 243 metabolites, the pH of the medium dropped from 7.6 to 4.7. Based on an average 244 biogas H₂ content of 68.6 \pm 1.4% (Fig. 1.A) it was calculated that 107.6 \pm 6.2 245 mL_{H2} (Fig. 1.B) and 73.9 \pm 6.2 mL_{CO2} were produced, yielding 1.10 \pm 0.05 mol 246 H₂/mol glucose (Fig. 1.C) at a maximum production rate of 0.98 \pm 0.04 mL_{H2}/h 247 248 (Fig. 1.D).

This H_2 yield is similar to that reported elsewhere in similar conditions [16] and

was only about half of that recorded with the same strain and medium in 2.3L

batch bioreactors, but at a fixed optimum pH of 5.2 [14]. The large difference

can be explained by the absence of pH regulation (ΔpH up to 3 units) and an 252 253 increase in pressure in the gas phase (up to 0.5 bar) during growth in BHP tests. The determination of the optimal concentration of casein peptone for hydrogen 254 255 production was carried out by testing 10 concentrations ranging from 0.025 to 2.3 g_N/L . With the sample containing twice the nitrogen content of the control 256 sample, the cumulative H₂ production, H₂ yields and production rates decreased 257 258 by 14%, 20%, and 21%, respectively. The trend was confirmed with a further increase of concentration up to 2.3 g_N/L (Fig. 1). 259

260 Since casein peptone concentrations higher than $0.56 g_N/L$ were not beneficial to 261 hydrogen production, the nitrogen content was decreased aiming to determine a lower concentration limit. A decrease from 0.56 to 0.25 g_N/L had a significant 262 positive impact on the H_2 yields and production rates (Fig. 1.C and 1.D). 263 Surprisingly this trend continued until the nitrogen content was reduced to 0.062 264 g_N/L . This 8-fold decrease resulted in a 30% increase in the H₂ yield compared 265 266 to the control sample (Fig. 1.C, from 1.10 ± 0.05 for the control sample to 1.43 $\pm 0.08 \text{ mol } H_2/\text{mol glucose}$). 267

However below $0.062 \text{ g}_{N}/\text{L}$ a very strong decrease occurred for both the H₂ yield and the H₂ production rate: 27% and 38% respectively (Fig. 1.C and 1.D). This shift may have been due to a lack of sufficient available nitrogen for cellular anabolism and basal metabolism [9]. This assumption was supported by the detection of increasing residual glucose at lower concentrations of casein peptone (Table 4, column 2). Therefore, as suggested in the literature [9, 28, 29],
the uptake of the substrate seems to be highly dependent on the nitrogen content
of the medium.

276 Based on these results the lower limit for optimized hydrogen production by Clostridium butyricum CWBI1009 was determined to be ca. 0.062 g_N/L of 277 organic nitrogen. This finding is highly relevant for cost reduction in biotech 278 279 applications, comparing to control or other conditions with similar peptonebased nitrogen source. Other opportunities to reduce cost could be investigated 280 281 through less expensive mineral nitrogen source or cheap residual organic source 282 providing the most essential amino acids for hydrogen production metabolism. The experiments reported below deal with culture medium containing 283 284 ammonium or amino acids and further analyses to assess their specific consumption by the *Clostridium* strain. 285

286

287 Bacto-tryptone and bacto-casamino acids as nitrogen source

To determine the effect of the complexity of the organic nitrogen source, tests were also carried out with easily assimilated pure nitrogen sources, namely bacto-tryptone (small peptides resulting from casein pancreatic digest) and bacto-casamino acids (mostly free amino acids), at 6 different concentrations (ranging from 0.56 to 0.056 g_N/L , see Supplementary Figure). The performances were compared to that obtained for equivalent conditions with casein peptone. 294 Under all the tested conditions the biogas H_2 content was on average 10% higher and a similar increase in H_2 yields was also recorded. In addition, the H_2 295 production rates were much higher than those achieved for the equivalent 296 297 nitrogen concentration of casein peptone. However, since the H₂ yields were very similar to those for the control sample with casein peptone (*i.e.* around 1 298 mol H₂/mol glucose) and considering the extremely high price of these pure 299 300 nitrogen sources, they were not investigated further. Interestingly the decrease in H₂ production activity was observed at exactly the same low concentration as 301 302 with the casein peptone source. These results enable to consider an optimal 303 organic nitrogen content of ca. $0.062 \text{ g}_{\text{N}}/\text{L}$.

304

305 *Ammonium sulfate as nitrogen source*

The prospect of using an ammonia-based substitute is of particular interest for 306 biotech applications since the cost of the nitrogen source could radically be 307 308 about 10-fold lower comparing to casein peptone. Therefore, we studied the effect of 10 different (NH₄)₂SO₄ concentrations (Fig. 1) used as a replacement 309 nitrogen source. Compared to the control sample the H₂ production rates 310 311 decreased by about 35% (Fig. 1.D). This observation was in line with the results of Liu et al. [30] which showed a correlation between a prolonged lag phase and 312 313 increasing $(NH_4)_2SO_4$ concentrations. However, the H₂ yields obtained with the $(NH_4)_2SO_4$ nitrogen source (*i.e.* a maximum of 1.11 ±0.04 mol H₂/mol glucose 314

at 0.25 g_N/L) were comparable to those for the control sample (Fig. 1.C). These trends for H₂ production yields and rates are consistent with the global benefit reported by Whang *et al.* [12] in continuous bioreactor fed by glucose and either peptone or ammonium nitrogen source. Analysis of the H₂ production activity showed that the inhibition pattern described above for low and high organic nitrogen contents was also observed with (NH₄)₂SO₄ (Fig. 1).

It has been shown that ammonia at high concentrations can inhibit fermentative hydrogen production, probably by changing the intracellular pH of the bacterium, increasing maintenance energy requirements or inhibiting specific enzymes such as the hydrogenase [9, 29-31].

In line with Liu *et al.* [30] we found that at high concentrations (*e.g.* 2.3 g_N/L) ammonia strongly inhibited fermentative H₂ production. By contrast Salerno *et al.* [29] found that the hydrogen yield was not affected by ammonia and averaged 1.07 ±0.04 mol H₂/mol glucose at 10 g_N/L ammonia. However they used a heat-treated sludge as inoculum that might involve microorganisms decreasing the inhibition effect caused by ammonia.

On the contrary, in the present study a pure isolate of *Clostridium butyricum* was used and the H₂ yield decreased when the ammonia concentration was outside a given range (*i.e.* from 0.56 to 0.062 g_N/L). This further demonstrated in accordance with Bisaillon *et al.* [31], Liu *et al.* [30] and Wang *et al.* [32] that decreasing the ammonia concentration could improve the H₂ yield. However, it should be noted that widely differing optimal ammonia concentrations have been reported for dark fermentation in the literature i.e. from 0.01 g_N/L reported by Bisaillon *et al.* [31] up to 10 g_N/L by Salerno *et al.* [29]. This can be explained by the very large differences between the studies regarding the seed sludge and substrate used.

341

342 Impact of carbon-to-nitrogen ratio and glucose uptake

In our results, an optimal 0.062 g_N/L of organic nitrogen was evidenced which 343 344 lead to about 50% higher H_2 yield than with the same nitrogen content of 345 ammonium sulfate and 30% higher H₂ yield than at the optimal ammonium concentration i.e. at a 4-fold higher nitrogen concentration leading to a H₂ 346 347 production rate about 25% lower. Regarding the related carbon-to-nitrogen ratio (C/N) that enables easier comparisons of results achieved in different 348 experimental conditions, the 29.3 value corresponding to the optimum 0.062 349 350 g_N/L is similar to that calculated from Cheong and Hansen [33] at optimal 0.8 g_N/L with ammonium nitrogen source. However, they did not record a 351 352 significant difference in biohydrogen production compared with experiments 353 with 2 g_N/L (i.e. C/N = 12). This trend is consistent with the results reported in our experiments with ammonium content varying from 0.062 to 0.56 g_N/L (i.e. 354 355 C/N in the range 3 - 30) and those of Wang *et al.* [32] with C/N in the range 8 -80. Lin and Lay [34], also using a simple carbon source (glucose) in batch 356

experiments, recorded an optimum C/N ratio of 47 within the range 40 -130. By contrast to our study, their H₂ production yields and rates decreased severely (>50%) for a slightly lower C/N of 40. Moreover, only H₂ yield decreased with C/N ratio increasing upto 98. Indeed, they report about a 2-fold decrease that is far less than the differences recorded in our results with C/N upto 72 for both H₂ production yields and rates (e.g. a 5-fold decrease for H₂ production rate by comparison to that at the optimum level).

Table 5 (column 2) shows the effect of the $(NH_4)_2SO_4$ on the substrate uptake efficiency. In line with Wang *et al.* [32] the glucose uptake rate was inhibited when ammonia was present as the sole nitrogen source. Furthermore, as with casein peptone, we observed a strong increase of glucose accumulation when the $(NH_4)_2SO_4$ content decreased from 0.062 g_N/L to 0.056 g_N/L. This was associated with a 37% reduction and an 11% reduction in the H₂ yield and production rate, respectively.

371

372 3.2. Soluble metabolite production

Tables 4 and 5 summarise the effect of casein peptone and $(NH_4)_2SO_4$ on the synthesis of soluble metabolites. The analysis was carried out by HPLC-RID and showed that the main metabolites produced by *Clostridium butyricum* CWBI1009 in the control sample were formate (14.39 ±1.96 mM), butyrate (12.91 ±2.27 mM) and acetate (9.45 ±1.30 mM). Compared to cultures in bioreactors [14], more reduced products such as lactate $(3.51 \pm 1.41 \text{ mM})$ and ethanol $(1.47 \pm 0.31 \text{ mM})$ were detected.

As suggested by Liu *et al.* [35]. Masset *et al.*[14] and Lin *et al.* [24] this could 380 381 be due to both the absence of pH regulation and the increasing hydrogen supersaturation in the liquid phase that are inherent characteristic of BHP batch 382 tests. When using 0.062 g_N/L casein peptone (the concentration at which the 383 384 maximum H₂ yield of 1.43 ± 0.08 mol H₂/mol glucose was obtained), the increased H₂ yield compared to the control sample could not be explained as the 385 386 result of actetate- and/or butyrate-type fermentation, since both acetate and 387 butyrate final concentrations were significantly lower than in control sample [14, 24]. By contrast the 50% decrease in the level of formate would suggest that 388 389 other metabolic pathways contributed to the H_2 yield improvement [20]. For the test condition with a casein peptone concentration of 0.056 g_N/L , a sharp 390 decrease in H₂ production activity was observed. This is in line with the findings 391 392 of Lin et al. [24], since an increase in lactate and ethanol production and a decrease in butyrate-type fermentation were also observed. 393

By contrast, the presence of $0.56 \text{ g}_{\text{N}}/\text{L}$ (NH₄)₂SO₄ favours butyrate production. This is consistent with the results reported by Wang *et al.* [28] and Salerno *et al.* [29]. The increase in butyrate implies that there was a gradual shift from acetateto butyrate-type fermentation [24]. Additionally formate production remained high at all (NH₄)₂SO₄ concentrations and did not correlate with the amount of residual glucose detected, suggesting that the increased formate production could be associated with a decrease in the activity of formate-tetrahydrofolate ligase which catalyses the initial uptake of C_1 units (one carbon units, [19]).

402 Compared to other studies with the *Clostridium butyricum* W5 isolate [28, 36], Clostridium butyricum CWBI1009 did not produce large amounts of lactate 403 under any of the conditions tested. Furthermore in other studies [14] we showed 404 405 that our isolate further converted lactate to H_2 (2 mol of lactate + 1 mol of acetate leading to butyrate formation + an additional 1 mol of H₂) and concluded 406 407 that *Clostridium butyricum* CWBI1009 has better potential than the W5 isolate 408 for future biotech applications. Interestingly this trend is confirmed when using an ammonia-based substitute since our isolate produced similarly low amounts 409 410 of lactate whether growing with a casein peptone or an ammonia-based nitrogen 411 source.

Out of the 20 test conditions described in the previous section, four were 412 413 selected for further analysis in order to better understand the differences observed in H_2 production: (i) the control sample with casein peptone (0.56 414 g_N/L), (ii) the sample providing the highest H₂ vield (0.062 g_N/L casein 415 416 peptone), (iii) the sample associated with the sharpest drop in the H₂ production rate (0.056 g_N/L casein peptone) and finally (iv) an ammonia-based sample with 417 a nitrogen content equivalent to that of the control sample (0.56 g_N/L 418 $(NH_4)_2SO_4$). The study consisted of a comparison between the control sample 419

and each of the last three samples. This was done by quantifying the amino acids
present in the culture supernatants (HPLC-UV) and by analysing the gene
expression patterns (RT-qPCR) for the four [FeFe]-hydrogenases and the
nitrogenase of *C. butyricum* CWBI1009.

424 **3.3. Amino acid distribution**

The quantitative analysis (HPLC-UV) of 20 proteogenic amino acids in the 425 426 culture supernatants was carried out initially and after 96 h of fermentation. The comparison of the control sample $(0.56 g_N/L \text{ of casein peptone})$ with the sample 427 associated with the highest H₂ yield (0.062 g_N/L casein peptone) confirm that in 428 429 the control sample only a small fraction of the available nitrogen was assimilated and, as a consequence, that the higher level of available nitrogen did 430 not favour high H₂ yields. This was also confirmed by another analysis which 431 showed that only 15% of the available organic nitrogen was incorporated into 432 the biomass (results not shown). At 0.062 g_N/L much more efficient amino acid 433 434 assimilation (+5.73 mM amino acids uptake compared to the control sample, Fig. 2.A) was observed with the strongest incorporation for MET, ARG and 435 GLY. 436

437 Since the 0.056 g_N/L casein peptone sample and the control sample have very 438 similar distribution patterns in Fig. 2.A and Fig. 2.B, it can be assumed that the 439 sharp 50% decrease in production rates recorded with the lower casein peptone 440 concentration could not be explained by a significant difference in the

441 assimilation of a specific amino acid. Moreover the average 35% decrease in H_2 442 production rates observed with the (NH₄)₂SO₄ sample would be related to the 443 high ATP cost of *de novo* glutamate biosynthesis to fix ammonia in the 444 GS/GOGAT cycle [29].

Interestingly the amino-acid profile for the ammonia-based sample (Fig. 2.C) 445 clearly indicated that C. butyricum CWBI1009 is capable of biosynthetising 446 447 (mainly ARG, MET, ALA) and fermenting (mainly CYS, ASN, LEU) proteogenic amino acids. This characteristic has been demonstrated for other 448 449 *Clostridium* strains (e.g. C. pasteurianum), but experimental evidence was 450 previously lacking for C. butyricum. It should be noted that the high apparent 451 incorporation of CYS indicated by the results in Fig. 2 could also be related to the addition of pure L-CYS at 0.5 g/L to provide the required anaerobic 452 conditions before inoculation. 453

454 **3.4.** [FeFe]-hydrogenase and nitrogenase expression patterns

It is now recognised that molecular microbial ecology tools can greatly facilitate scientific investigation of the processes involved in environmental biotechnology. Therefore Real-Time Quantitative PCR was used for the four [FeFe]-hydrogenases and the nitrogenase (nifH subunit) identified in the genome of *C. butyricum* to provide a deeper insight into the role played by these two enzyme groups [19]. 461 To correlate the expression patterns with an increase or a decrease in H₂ 462 production activity we compared the control sample ($0.56 \text{ g}_N/\text{L}$ casein peptone) 463 with the other three above-mentioned conditions (namely, $0.062 \text{ g}_N/\text{L}$ and 0.056464 g_N/L casein peptone and $0.56 \text{ g}_N/\text{L}$ (NH₄)₂SO₄).

With the 0.062 g_N/L casein peptone sample the highest H₂ yield (1.43 ±0.08 mol 465 H_2 /mol glucose) occurred with up-regulation of the *hvdB2* and *hvdB3* 466 467 hydrogenase genes (Fig. 3.A). However the basal expression level of hydB2 (measured as the number of cDNA copies per 1000 cDNA copies of 16S rRNA, 468 469 Fig. 4) was higher than that of hydB3, which suggests that the increased H₂ yield 470 can be attributed to the activity of the HydB2 enzyme. This is confirmed by the similar basal expression level of the *hydB2* hydrogenase gene (Fig. 4) recorded 471 472 in the control sample and in the experiment with 0.056 g_N/L that led to a sharp decrease in H_2 production activity (Fig. 1). 473

This assumption about the key role of the HydB2 enzyme was also supported by another study comparing the expression patterns of the *Clostridium butyricum* CWBI1009 hydrogenase genes at the optimal pH value for growth (pH 7.3) and for H₂ production (pH 5.2) [19]. In that study we also suggested that the increased H₂ production at low pH could be attributed to the HydB2 enzyme.

479 Regarding *hydA2* gene that has the closest identity match with *Clostridium* 480 *pasteurianum* CP1, the results of the present study show similar expression 481 levels under the control condition and both the 0.056 g_N/L casein peptone and 482 $0.56 \text{ g}_{N}/\text{L} (\text{NH}_{4})_2 \text{SO}_4$ test conditions. By contrast, under the 0.062 g_{N}/L casein 483 peptone condition, *hydA2* was slightly down-regulated (Fig. 3.A) whereas its 484 basal expression level was higher than that of the other three hydrogenase genes 485 (Fig. 4).

The differences observed in the relative expression levels in the three test 486 samples and the control sample (Fig. 3) show that the nitrogenase gene was very 487 488 sharply down-regulated with 0.062 g_N/L peptone and 0.56 g_N/L (NH₄)₂SO₄, and at a lower extent with 0.056 g_N/L casein peptone. This suggests that with 0.062 489 490 g_N/L casein peptone H₂ production was essentially hydrogenase mediated. The 491 quite high basal expression level recorded for the nitrogenase in the control 492 sample (Fig. 4), while the H₂ yield was lower than with 0.062 g_N/L casein 493 peptone, suggests that nitrogenase-mediated H_2 production by *Clostridium* butyricum CWBI1009 may be less efficient than hydrogenase-mediated H₂ 494 production. 495

496

497 4. Conclusion

This research aimed to investigate fermentative hydrogen production by *Clostridium butyricum* CWBI1009 with different nitrogen sources. The results show increasing glucose accumulation when casein peptone or $(NH_4)_2SO_4$ concentration decreased below 0.062 g_N/L. This concentration of peptone leads to the optimal overall performance *i.e.* an H₂ yield and production rate of 1.43

mol H₂/mol glucose and 1.08 mL_{H2}/h, respectively. The amino acid distribution 503 504 and the gene expression pattern for hydrogenases and nitrogenase confirm that, 505 at a higher nitrogen concentration, there was an excess of nitrogen. These results 506 are promising for industrial implementation of the bioprocess. They enable, based on the carbon and nitrogen content of the main substrates, to futher 507 determine the optimized conditions for operating dark fermentation in (semi-508 509)continuous mode leading to minimum residues and maximum H₂ production yields and rates. They also confirm the key role of HydB2 in the optimisation of 510 hydrogen production by clostridia. Further investigations of this specific enzyme 511 512 in larger scale bioreactors and especially in biofilm bioreactors providing better H_2 gas transfer should be encouraged. 513

514

515 Acknowledgements

This work, C. Hamilton and M. Calusinska were supported by an ARC project (Action de Recherches Concertées, ULg-ARC 07/12 04). J. Masset and L. Beckers were recipients of a FRIA and FRS-FNRS fellowship, respectively (Fonds de la Communauté française de Belgique pour la Formation à la Recherche dans l'Industrie et l'Agriculture, Fonds de la Recherche Scientifique).

522

523 **References**

- 524 [1] Marbán G, Valdés-Solís T. Towards the hydrogen economy? Int J Hydrogen
 525 Energy 2007;3212:1625-1637.
- 526 [2] Munir R, Levin DB. Enzyme Systems of Anaerobes for Biomass
 527 Conversion. In: Hatti-Kaul R, Mamo G, Mattiasson B, editors.
 528 Anaerobes in Biotechnology, Switzerland: Springer International
 529 Publishing; 2016, p. 113-138.
- [3] Hallenbeck PC, Abo-Hashesh M, Ghosh D. Strategies for improving
 biological hydrogen production. Bioresour Technol 2012;110:1-9.
- 532 [4] Garcia Depraect O, Gomez-Romero J, Leon-Becerril E, Lopez-Lopez A. A
- novel biohydrogen production process: Co-digestion of vinasse and *Nejayote* as complex raw substrates using a robust inoculum. Int J
 Hydrogen Energy 2017;42:5820-5831.
- [5] Mohan SV, Mohanakrishna G, Srikanth S. Biohydrogen Production from
 Industrial Effluents. In: Pandey A, Larroche C, Ricke SC, Dussap CG,
 Gnansounou E. Biofuels, Amsterdam: Academic Press; 2011, p. 499524.
- [6] Sydney EB, Larroche C, Novak AC, Nouaille R, Sarma SJ, Brar SK, Letti
 LAJ, Soccol VT, Soccol CR. Economic process to produce biohydrogen
 and volatile fatty acids by a mixed culture using vinasse from sugarcane
 ethanol industry as nutrient source. Bioresour Technol 2014;159:380–
 386.

545	[7] Wang X, Ding J, Guo WQ, Ren NQ. Scale-up and optimisation of
546	biohydrogen production reactor from laboratory-scale to industrial-scale
547	on the basis of computational fluid dynamics simulation. Int J Hydrogen
548	Energy 2010;3520:10960-10966.
549	[8] Guo XM, Trably E, Latrille E, Carrère H, Steyer JP. Hydrogen production
550	from agricultural waste by dark fermentation: A review. Int J Hydrogen
551	Energy 2010;3519:10660-10673.
552	[9] Amon J, Titgemeyer F, Burkovski A. Common patterns – unique features:
553	nitrogen metabolism and regulation in Gram-positive bacteria. FEMS
554	Microbiol Rev 2010;344:588-605.
555	[10] Wang J, Wan W. Factors influencing fermentative hydrogen production: A
556	review. Int J Hydrogen Energy 2009;34:799-811.
557	[11] Hamilton C, Hiligsmann S, Beckers L, Masset J, Wilmotte A, Thonart P.
558	Optimisation of culture conditions for biological hydrogen production by
559	Citrobacter freundii CWBI952 in batch, sequenced-batch and
560	semicontinuous operating mode. Int J Hydrogen Energy 2010;353:1089-
561	1098.
562	[12] Whang LM, Lin CA, Wu CW, Cheng HH. Metabolic and energetic aspects
563	of biohydrogen production of Clostridium tyrobutyricum: The effects of
564	hydraulic retention time and peptone addition. Bioresour Technol
565	2011;102:8378-8383

566	[13] Fernandes BS, Peixoto G, Albrecht FR, Saavedra del Aguila NK, Zaiat M.					
567	Potential to produce biohydrogen from various wastewaters. Energy					
568	Sustain Develop 2010;142:143-148.					
569	[14] Masset J, Hamilton C, Hiligsmann S, Beckers L, Franck F, Thonart P.					
570	Effect of pH on glucose and starch fermentation in batch and sequenced-					
571	batch mode with a recently isolated strain of hydrogen-producing					
572	Clostridium butyricum CWBI1009. Int J Hydrogen Energy					
573	2010;358:3371-3378.					
574	[15] Masset J, Calusinska M, Hamilton C, Hiligsmann S, Joris B, Wilmotte A,					
575	Thonart P. Fermentative hydrogen production from glucose and starch					
576	using pure strains and artificial co-cultures of Clostridium spp.					
577	Biotechnol Biofuels 2012;5:1-15.					
578	[16] Hiligsmann S, Masset J, Hamilton C, Beckers L, Thonart P. Comparative					
579	study of biological hydrogen production by pure strains and consortia of					
580	facultative and strict anaerobic bacteria. Bioresour Technol					
581	2011;1024:3810-3818.					
582	[17] Hiligsmann S, Beckers L, Masset J, Hamilton C, Thonart P. Improvement					
583	of fermentative biohydrogen production by Clostridium butyricum					
584	CWBI1009 in sequenced-batch, horizontal fixed bed and biodisc					
585	anaerobic reactors with biomass retention. Int J Hydrogen Energy					
586	2014,39:6899-6911.					

587	[18] Soboh B, Linder D, Hedderich R. A multisubunit membrane-bound [NiFe]-
588	hydrogenase and an NADH-dependent Fe-only hydrogenase in the
589	fermenting bacterium Thermoanaerobacter tengcongensis. Microbiol
590	2004;1507:2451-2463.
591	[19] Calusinska M, Hamilton C, Monsieurs P, Mathy G, Leys N, Franck F, Joris
592	B, Thonart P, Hiligsmann S, Wilmotte A. Genome-wide transcriptional
593	analysis suggests hydrogenase- and nitrogenase-mediated hydrogen
594	production in Clostridium butyricum CWBI 1009. Biotechnol Biofuels
595	2015;8:1-16.
596	[20] Calusinska M, Happe T, Joris B, Wilmotte A. The surprising diversity of
597	clostridial hydrogenases: a comparative genomic perspective. Microbiol
598	2010;1566:1575-1588.
599	[21] Mulder DW, Shepard EM, Meuser JE, Joshi N, King PW, Posewitz MC,
600	Broderick JB, Peters JW. Insights into [FeFe]-Hydrogenase Structure,
601	Mechanism, and Maturation. Struct 2011;198:1038-1052.
602	[22] Calusinska M, Joris B, Wilmotte A. Genetic diversity and amplification of
603	different clostridial [FeFe]-hydrogenases by group-specific degenerate
604	primers. Lett in Appl Microbiol 2011;534:473-480.
605	[23] Savichtcheva O, Joris B, Wilmotte A, Calusinska M. Novel FISH and
606	quantitative PCR protocols to monitor artificial consortia composed of

607 different hydrogen-producing *Clostridium spp*. Int J Hydrogen Energy
608 2010;3613:7530-7542.

609 [24] Lin PY, Whang LM, Wu YR, Ren WJ, Hsiao CJ, Li SL, Chang JS.

- Biological hydrogen production of the genus *Clostridium*: Metabolic
 study and mathematical model simulation. Int J Hydrogen Energy
 2007;32: 1728-1735.
- [25] Rasmussen R. Quantification on the lightcycler. In: Rapid cycle real-time
 PCR, methods and applications, New York: Springer Press; 2000, p. 21–
 34.
- [26] Pfaffl MW, Horgan GW, Dempfle L. Relative expression software tool
 REST© for group-wise comparison and statistical analysis of relative
 expression results in real-time PCR. Nucleic Acids Res 2002; 30:e36.
- [27] Pfaffl MW, Tichopad A, Prgomet C, Neuvians TP. Determination of stable
 housekeeping genes, differentially regulated target genes and sample
 integrity: BestKeeper Excel-based tool using pair-wise correlations.
- 622 Biotechnol Lett 2004;266:509-515.
- [28] Wang X, Jin B, Mulcahy D. Impact of carbon and nitrogen sources on
 hydrogen production by a newly isolated *Clostridium butyricum* W5. Int
 J Hydrogen Energy 2008;3319:4998-5005.
- [29] Salerno MB, Park W, Zuo Y, Logan BE. Inhibition of biohydrogen
 production by ammonia. Water Res 2006;406:1167-1172.

628	[30] Liu G, Shen J. Effects of culture and medium conditions on hydrogen
629	production from starch using anaerobic bacteria. J Biosci Bioeng
630	2004;984:251-256.
631	[31] Bisaillon A, Turcot J, Hallenbeck PC. The effect of nutrient limitation on
632	hydrogen production by batch cultures of Escherichia coli. Int J
633	Hydrogen Energy 2006;3111:1504-1508.
634	[32] Wang B, Wan W, Wang J. Effect of ammonia concentration on
635	fermentative hydrogen production by mixed cultures. Bioresour Technol
636	2009;1003:1211-1213.
637	[33] Cheong DY, Hansen CL. Feasibility of hydrogen production in
638	thermophilic mixed fermentation by natural anaerobes. Bioresour
639	Technol 2007;98:2229–2239.
640	[34] Lin CY, Lay CH. Carbon/nitrogen-ratio effect on fermentative hydrogen
641	production by mixed microflora. Int J Hydrogen Energy 2004;291:41-
642	45.
643	[35] Liu IC, Whang LM, Ren WJ, Lin PY. The effect of pH on the production
644	of biohydrogen by clostridia: Thermodynamic and metabolic
645	considerations. Int J Hydrogen Energy 2011;361:439-449.
646	[36] Cai G, Jin B, Saint CP, Monis PT. Metabolic flux analysis of hydrogen
647	production network by Clostridium butyricum W5: Effect of pH and
648	glucose concentrations. Int J Hydrogen Energy 2010;3513:6681-6690.

649	
650	
651	
652	
653	
654	
655	Table captions
656	Table 1 : Injection and derivatisation program for HPLC-UV analysis of amino
657	acids.
658	
659 660	Table 2 : Mobile phase composition and gradient conditions for HPLC-UV analysis of amino acids.
661	
662	Table 3 : Specific primers and probes used in this study. 1 AT – PCR
663	annealing temperature. ² HP – hydrolysis probe.
664	
665	Table 4 : Glucose conversion and distribution of soluble metabolites produced
666	by Clostridium butyricum CWBI1009 after 96 h of culture in 270 mL serum
667	vessels with casein peptone as nitrogen source. SD, Standard Deviation on
668	triplicates; N.D., No data available.

670	Table 5 : Glucose conversion and distribution of soluble metabolites produced
671	by Clostridium butyricum CWBI1009 after 144 h of culture in 270 mL serum
672	vessels with $(NH_4)_2SO_4$ as nitrogen source. SD, Standard Deviation on
673	triplicates.
674	
675	
676	
677	
678	Figure captions
679	
680	Figure 1 : Clostridium butyricum CWBI1009 grown with 5 g/L glucose in 270
681	mL batch bottles with 10 different concentrations of casein peptone and
682	$(NH_4)_2SO_4$ expressed in terms of total nitrogen content. [A] Biogas H ₂ content,
683	[B] Cumulative H_2 production, [C] H_2 yield, [D] H_2 production rate. At least
684	three biological replicates were performed for each condition. The control
685	sample is indicated by a black arrow.
686	
687	Figure 2 : Amino acid distribution, compared to the control sample (0.56 g_{N}/L
688	casein peptone), in the culture supernatant of Clostridium butyricum CWBI1009
689	grown with 5 g/L glucose in 270 mL batch cultures. [A] 0.062 g_{N}/L casein
690	peptone, [B] 0.056 g_N/L casein peptone and [C] 0.56 g_N/L (NH ₄) ₂ SO ₄ . The

- samples were derivatised online and analysed by HPLC-UV at 262 nm forhydroxyproline and proline and at 338 nm for the other amino acids.
- 693

694 Figure 3 : Differential expression of [FeFe]-hydrogenase genes (*hydA2*, *hydA8*, hydB2 and hydB3) and the nitrogenase gene (nifH) of Clostridium butyricum 695 CWBI1009 and comparison of the gene expression patterns. [A] 0.062 g_N/L and 696 697 [B] 0.056 g_N/L casein peptone and [C] 0.56 g_N/L (NH₄)₂SO₄ compared to 0.56 g_N/L casein peptone (control sample). The standard bar graph visualisation was 698 699 replaced by a statistical whisker-box plot to provide additional information 700 about the skew of the distributions that would not be available by simply plotting the sample mean. *Statistically not different from the control sample, 701

p>0.05.

Figure 4 : Basal expression of [FeFe]-hydrogenase genes and the nitrogenase
gene of *Clostridium butyricum* CWBI1009 with casein peptone and (NH₄)₂SO₄.
The basal expression level is shown as the number of cDNA copies per 1000
cDNA copies of 16S rRNA.

708

Supplementary Figure 1 : *Clostridium butyricum* CWBI1009 grown with 5 g/L
glucose in 270 mL batch bottles with different concentrations of bacto-tryptone
and bacto-casamino acids, expressed in terms of the total nitrogen. [A] Biogas

- H_2 content, [B] Cumulative H_2 production, [C] H_2 yield, [D] H_2 production rate.
- 713 At least three biological replicates were performed for each condition.

Line	Function	Amount	Reagent
1	borate buffer		
2	draw	0.5 μL	Sample
3	mix	$3.0 \ \mu$ L in air, $400 \ \mu$ L/min, 2 times	
4	wait	0.50 min	
5	draw	0.0 μL	Water-needle wash
6	draw	1.0 μL	OPA-3MPA
7	mix	3.5 μ L in air, 400 μ L/min, 6 times	
8	draw	0.0 μL	Water-needle wash
9	draw	0.5 μL	FMOC
10	mix	$4.0 \ \mu$ L in air, $400 \ \mu$ L/min, 6 times	
11	draw	32.0 µL	water
12	mix	18.0 μ L in air, 400 μ L/min, 2 times	
13	inject	(0.5 µL)	

Time	Mobile phase A	Mobile phase B
		(45% acetonitrile, 45%
(min)	(40 mmol/l Na ₂ HPO ₄ , pH 7.8)	methanol, 10% water)
0.00	100.0	0.0
1.90	100.0	0.0
8.00	79.7	21.3
19.30	44.8	55.2
19.80	0.0	100.0
23.50	0.0	100.0
24.40	100.0	0.0
25.20	100.0	0.0

Target gene	Primer name	Sequence $5' \rightarrow 3'$	Length of product (bp)	AT ¹ (°C)
recA	RecA-butF RecA-butR	AAGCATTAGTGCGTTCTGGAG GAATCTCCCATTTCCCCTTC	97	60
gyrA	GyrA-butF GyrA-butR	AGCAATGGGTAGAACTGCATC ATTCTTCGCCATCAACTGCT	95	60
hydA2	ButA2F ButA2R	ATAGTTGCAATGGCTCCTGC TTTCTGCTTGCCTAACCCAT	250	60
hydA8	ButA8F ButA8R	TCTTTGGAGTTACAGGGGGA TTCAGCATTTGCAAGACCAC	188	60
hydB2	ButB2F ButB2R	TGGTGGTGTATCAACTGCTG TTGCATCCCATTCCTTCAAT	168	60
hydB3	ButB3F ButB3R	CAATGGTTGCTACAGGCAGA CAAAAGCATCGAATAACGCA	168	60
16S rRNA	16SButF 16SbutR HP16Sbut ²	CCTGCCTCATAGAGGGGAAT GAGCCGTTACCTCACCAACT CCGCATAAGATTGTAGTACCG CATGGTACA	143	60

Casein peptone						
concentration	Glucose	Lactate	Formate	Acetate	Ethanol	Butyrate
(g_N/L)	$mmol \ \pm \ SD$					
2.3	$4.93 \hspace{0.2cm} \pm \hspace{0.2cm} 1.44$	$6.79 \hspace{0.2cm} \pm \hspace{0.2cm} 3.91$	1.25 ± 0.46	$0.87 ~\pm~ 0.29$	$5.18 \hspace{0.2cm} \pm \hspace{0.2cm} 0.97$	$4.64 \hspace{0.2cm} \pm \hspace{0.2cm} 1.18$
1.3	$3.64 \ \pm \ 0.93$	$4.95 ~\pm~ 1.01$	1.63 ± 1.01	1.08 ± 0.53	$4.81 \hspace{0.2cm} \pm \hspace{0.2cm} 1.09$	$4.77 \hspace{0.2cm} \pm \hspace{0.2cm} 2.19$
0.56	$0.44 \hspace{0.2cm} \pm \hspace{0.2cm} 0.09$	$3.51 ~\pm~ 1.41$	$14.39 ~\pm~ 1.96$	$9.45 ~\pm~ 1.30$	1.47 ± 0.31	$12.91 \ \pm \ 2.27$
0.37	$0.11 \hspace{0.2cm} \pm \hspace{0.2cm} 0.07$	$3.86 ~\pm~ 0.82$	$18.66 ~\pm~ 1.64$	$5.61 \hspace{0.2cm} \pm \hspace{0.2cm} 1.64$	$0.15 \hspace{0.2cm} \pm \hspace{0.2cm} 0.08$	$13.32 \ \pm \ 1.15$
0.25	$0.70 \hspace{0.2cm} \pm \hspace{0.2cm} 0.23$	$3.91 ~\pm~ 0.33$	15.72 ± 2.52	$3.68 \hspace{0.2cm} \pm \hspace{0.2cm} 1.12$	$0.88 \hspace{0.2cm} \pm \hspace{0.2cm} 0.67$	$9.44 \hspace{0.2cm} \pm \hspace{0.2cm} 1.19$
0.12	$8.33 ~\pm~ 1.79$	$3.68 ~\pm~ 0.24$	$16.32 \ \pm \ 2.59$	$3.90 ~\pm~ 1.17$	$0.95 ~\pm~ 0.33$	$9.46 ~\pm~ 1.13$
0.062	$10.58 ~\pm~ 3.00$	$2.80 ~\pm~ 1.50$	$7.80 ~\pm~ 2.86$	$3.11 \hspace{.1in} \pm \hspace{.1in} 1.47$	$3.57 ~\pm~ 1.14$	$6.54 \hspace{0.2cm} \pm \hspace{0.2cm} 1.37$
0.056	$13.43 ~\pm~ 1.50$	$4.10 ~\pm~ 0.34$	$6.01 \hspace{0.2cm} \pm \hspace{0.2cm} 1.70$	$2.68 \hspace{0.2cm} \pm \hspace{0.2cm} 0.25$	$7.59 ~\pm~ 1.26$	$2.10 ~\pm~ 0.56$
0.037	N.D.	N.D.	N.D.	N.D.	N.D.	N.D.
0.025	$15.00 \ \pm \ 2.04$	$6.10 \hspace{0.1in} \pm \hspace{0.1in} 1.01$	$4.46 ~\pm~ 1.30$	$1.89 \ \pm \ 0.21$	$9.52 \hspace{0.2cm} \pm \hspace{0.2cm} 1.35$	$1.29 ~\pm~ 0.61$

$(NH_4)_2SO_4$						
Concentration	Glucose	Lactate	Formate	Acetate	Ethanol	Butyrate
(g_N/L)	$mmol \ \pm \ SD$	$mmol \ \pm \ SD$	$mmol \ \pm \ SD$	$mmol \ \pm \ SD$	$mmol \ \pm \ SD$	$mmol \ \pm \ SD$
2.3	$3.47 ~\pm~ 0.33$	$3.76~\pm~1.07$	$17.06 ~\pm~ 2.88$	$6.02 \hspace{0.2cm} \pm \hspace{0.2cm} 1.75$	$3.09 ~\pm~ 1.75$	$4.09 \hspace{0.2cm} \pm \hspace{0.2cm} 1.02$
1.3	$4.21 \hspace{0.2cm} \pm \hspace{0.2cm} 1.64$	$3.10~\pm~0.72$	$17.10 ~\pm~ 1.72$	$6.08 \hspace{0.2cm} \pm \hspace{0.2cm} 1.84$	$2.22 \ \pm \ 1.81$	$3.75 ~\pm~ 1.70$
0.56	$3.02 \hspace{0.2cm} \pm \hspace{0.2cm} 1.55$	$3.85 ~\pm~ 1.58$	$17.01 \hspace{0.1 in} \pm \hspace{0.1 in} 1.30$	$6.42 \hspace{0.2cm} \pm \hspace{0.2cm} 1.36$	$1.31 ~\pm~ 0.60$	$12.73 \ \pm \ 2.36$
0.37	$2.67 ~\pm~ 0.53$	$3.80 ~\pm~ 0.87$	$18.32 \ \pm \ 2.27$	$6.05 \hspace{0.2cm} \pm \hspace{0.2cm} 1.64$	1.72 ± 0.76	$13.80 ~\pm~ 1.62$
0.25	$2.33 ~\pm~ 0.51$	$3.55 ~\pm~ 0.63$	$17.74 \hspace{0.2cm} \pm \hspace{0.2cm} 1.17$	$6.82 \hspace{0.2cm} \pm \hspace{0.2cm} 2.17$	$1.13 ~\pm~ 0.33$	$12.06 ~\pm~ 1.98$
0.12	$2.21 \hspace{.1in} \pm \hspace{.1in} 1.11$	$3.83 ~\pm~ 0.53$	$19.08 ~\pm~ 3.00$	$6.93 ~\pm~ 1.95$	$0.92 ~\pm~ 0.40$	$13.09 ~\pm~ 2.72$
0.062	$2.01 \hspace{0.2cm} \pm \hspace{0.2cm} 0.88$	3.68 ± 1.21	$18.93 ~\pm~ 1.52$	$7.34 ~\pm~ 1.39$	$1.39 ~\pm~ 0.54$	$13.32 \ \pm \ 1.29$
0.056	$8.13 ~\pm~ 1.40$	$4.07 \hspace{0.2cm} \pm \hspace{0.2cm} 2.02$	$13.50 ~\pm~ 1.92$	$6.83 \hspace{0.2cm} \pm \hspace{0.2cm} 1.34$	$2.12 ~\pm~ 0.50$	$2.65 ~\pm~ 0.99$
0.037	11.32 ± 2.56	1.52 ± 1.18	$10.52 \ \pm \ 1.61$	5.14 ± 1.68	$2.55 ~\pm~ 0.60$	$7.13 ~\pm~ 1.39$
0.025	$11.92 \ \pm \ 2.34$	$2.26 ~\pm~ 2.48$	$9.87 ~\pm~ 2.02$	$4.63 \hspace{0.2cm} \pm \hspace{0.2cm} 1.63$	$3.87 \ \pm \ 1.41$	$3.08 ~\pm~ 0.34$









